CONTRACT AF 33(615). 2400 PROJECT 7381; TASK 738101

AD 465161

SILICON CARBIDE

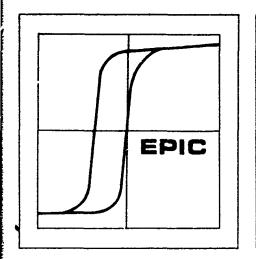
Data Sheets

M. Neuberger

Approved for public release; distribution is unlimited.

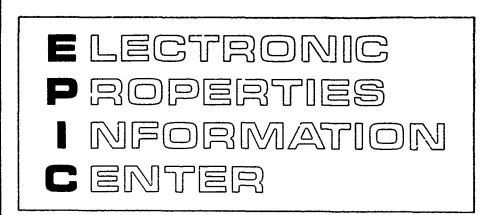
DS-145

June 1965



2

3-46516



HUGHES

HUGHES AIRCRAFT COMPANY CULVER CITY CALIFORNIA

NOTICES

When Government drawings, specifications, or other data are used for any purpose other than in connection with a definitely related Government procurement operation, the United States Government thereby incurs no responsibility nor any obligation whatsoever; and the fact that the Government may have formulated, furnished, or in any way supplied the said drawings, specifications, or other data, is not to be regarded by implication or otherwise as in any manner lipensing the holder or any other person or corporation, or conveying any rights or permission to manufacture, use, or sell any patented invention that may in any way be related thereto.

Qualified requesters may obtain copies of this report from the Defense Documentation Center (DDC), Cameron Station, Bldg. 5, 5010 like Street. Alexandria, Virginia, 22314. The distribution of this report is limited because the report contains technology identifiable with items on the strategic embargo lists excluded from export or re-export under U.S. Export Control Act of 1949 (6.8 STAT. 7), as amended (50 U.S.C. App. 2026, 2031), as implemented by MER 400-10.

Approved for public release; distribution is unlimited.

Copies of this report should not be returned to the Research and Technology Division, Wright-Patterson Air Force Base, Ohio, unless return is required by security considerations, contractural obligations, or notice on a specific document.

CONTRACT AF 33(615). 2480 PROJECT 7381; TASK 738103

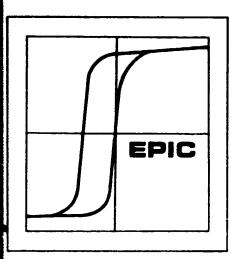
SILICON CARBIDE

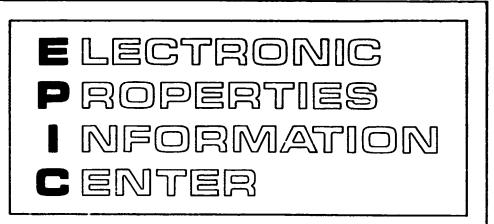
Data Sheets

M. Neuberger

DS-145

June 1965





HUGHES

HUGHES AIRCRAFT COMPANY CULVER CITY CALIFORNIA

FOREWORD

This report was prepared by Hughes Aircraft Company under Contract Mo. AF 33 (615)-2460. The contract was initiated under Project No. 7381, Task No. 738103. The work was administered under the direction of the Air Force Materials Laboratory, Research and Technology Division, with Mr. R. F. Klinger acting as Project Engineer.

The Electronic Properties Information Center has been established to collect, index and abstract the literature on the electrical and electronic properties of materials and to evaluate and compile the experimental data from that literature. A modified coordinate index to the literature is machine-stored and printed for manual use. The Center published summary reports, thesauri, glossaries, data sheets and similar publications as sufficient information is evaluated and compiled. This report consists of the compiled data sheets on Silicon Carbide.

Many persons have contributed to the program which this report represents. The author wishes especially to acknowledge the contributions of the following: John W. Atwood, C.L.M. Blocher, D. Gough, D.L. Grigsby, D.H. Johnson, H. Thayne Johnson, Thomas J. Lyndon, J.T. Milek, Emil Schafer and C.A. Schill.

Special thanks are extended to Dr. J. J. Grossman for his efforts in reviewing all technical data prior to publication.

ABSTRACT

These data sheets present a compilation of a wide range of electrical, optical and energy values for alpha and beta-silicon carbide in bulk and film form. Electrical properties include conductivity, resistivity, dielectric constant, Hall coefficient, mobility, lifetime and thermoelectric effects. Emission data have been broken down into the varied electron and photon emissions which result from application of energy in the electromagnetic spectrum. Energy data include energy bands, energy gap and energy levels for variously-doped silicon carbide, as well as effective mass tables, work function, and cross sections. The optical properties include absorption, reflection and refractive index. Other magnetic data and irradiation effects are included, as well as several bordering physical phenomena, such as piezoelectric properties, Debye temperature and thermal conductivity. Each property is compiled over the widest possible range of parameters from references obtained in a thorough literature search.

A crystallographic summary has been added.

This report has been reviewed and is approved for publication.

H. Thayne Johnson, Head

Electronic Properties Information Center

John W. Atwood Project Manager

TABLE OF CONTENTS

	Page
Foreword	.i11
Abstract	
Introduction	
Crystallography	
Absorption	9
Cross Sections	
Debye Temperature	
Dielectric Constant.	
Effective Mass	
Electrical Conductivity.	
Electrical Resistivity	
Electroacoustic Properties	
Electron Field Emission.	-
Electron Photoemission	
Electron Thermionic Emission	
Energy Bands	
Energy Gap	
Energy Levels	
Hall Coefficient	
Irradiation Properties	-
Lifetime	
Magnetic Susceptibility	
Magnetoelectric Properties	
Mobility	
Photoelectronic Properties	
Photon Electroluminescence	
Photon Emissivity	
Photon Luminescence	
Photon Thermoluminescence	
Reflectivity	
Refractive Index	
Thermal Conductivity	
Thermoelectric Properties	
Work Function	
References	. 97
Publications of the Electronic Properties Information Center	

INTRODUCTION

In June 1961, a program was initiated under the direction of the Air Force to collect, index and abstract the literature on the electrical and electronic properties of materials and to evaluate and compile the experimental data from that literature. Placed at Hughes Aircraft Company in Culver City, California, the program, now called the Electronic Properties Information Center, was originally intended to cover ten major categories of materials: Semiconductors, Insulators, Ceramics, Ferroelectrics, Metals, Ferrites, Ferromagnetics, Electroluminescent Materials, Thermionic Emitters, and Superconductors.

During the first year, studies were completed on the Semiconductor and Insulator categories; and Ceramics was discontinued as a separate category and subsumed under the other nine. Vocabulary studies have now been completed on all categories, and retrospective documentation is virtually complete for Semiconductors and Insulators. A full index to the literature is maintained, and publications such as data sheets, summary reviews, glossaries, and thesauri are issued periodically. The use of the Center and these publications are available to anyone wishing information within the scope of the Center's objectives. A full list of publications to date appears at the end of this report.

The first step in the preparation of data sheets is the retrieval, by manual modified coordinate index, of all literature on the material to be compiled. Bibliographies are checked to ensure the inclusion of all relevant literature and the assembled articles are given to the specialist doing the evaluation and compilation.

Evaluation is confined to primary source data except when only secondary citations are available. If equally valid data are available from several

sources, all are given. Data are rejected when judged questionable because of faulty or dubious measurements, unknown sample composition, or if more reliable data are available from another source. Selection of data is based upon that which is judged most representative, precise, reliable and which covers the widest range of variables.

Pure silicon carbide is colourless, with a Mohs hardness over 9, almost equal to that of the diamonds that Acheson was attempting to synthesize when he created silicon carbide artifically in the arc furnace in 1891. The natural material, Moissanite, was discovered as small hexagonal plates in meteoric iron about 15 years later in 1904. Commercial silicon carbide is black, with a bluish or greenish iridescence, it is refractory to heat up to 1000°C, unreactive to chemical attack and melts at 2830°C. It is now produced in large quantities by the electric furnace method at Niagara Falls, and is used extensively as an abrasive, for drills and highly refractory bricks. In 1963, its capability as a CW diode laser was disclosed, [Ref. 17972].

A silicon carbide p-n junction operating at room temperature has an emission peak at 4560Å with a linewidth of less than 5Å, at threshold current densities of 120 amps/cm². This important discovery has renewed interest in the material. It is however, an extremely difficult and refractory material to handle and this fact is reflected in the quantity of published material. One hundred and eighty four papers in all, were available to represent work done since about 1955. The older data are well covered in Gmelins Handbuch, and it is only the new techniques of the past ten years that have made possible samples of predetermined quality. Of the one hundred and eighty four papers, seventy nine were used in these data sheets.

Within every property section, we have tried to include every possible parameter and range of experimental condition tested. All the available information on test and sample specifications is also given. All graphs and tabular data are extracted from the article and quoted without alteration of any kind.

In previous compilations, materials have been arranged to show firstly data on pure samples, and then by the various dopants. Silicon carbide however, has thus far, not been produced pure from the standpoint of either dopants or crystal type. This question will be discussed more fully in the section on crystallography. In any event, the arrangement is practically self-explanatory, the data being divided into the alpha and beta forms, single crystal and films, and the nitrogen-doped (n-type) or the aluminum-doped (p-type).

In tabular data, values are variously arranged. In some cases it is by dopant, in others by numerical value. On occasion, however, the values from one reference may be kept together for simple comparison.

The references, from which the data are drawn, are shown by reference numbers below each graph with the full bibliographic information at the end of the data sheets. The bibliography is referred to and listed in the order of entry into the Center (accession number). This provides a quick cross reference into the index used with the literature.

CRYSTALLOGRAPHY

Silicon carbide is an unique material and a précis of its crystallographic characteristics is thoroughly covered by <u>Gmelins Handbuch</u>. It is of the utmost importance to note, however, that the energy gap value changes with crystal structure by 1 ev; from 2.2 ev for the beta-silicon carbide to 3.12 ev for the alpha-silicon carbide (6H). Such details, germane to the electronic properties of silicon carbide are given here.

Silicon carbide is both polymorphic and polytypic. The beta form is cubic and also called the low-temperature form. The alpha form is hexagonal and has 37 types of which 14 are hexagonal and 23 are rhombohedral. Many of these types are structurally equivalent in two directions with the same dimensions and lattice arrangement but dissimilar in the c-direction. Nomenclature is given according to Ramsdell, the chief investigator in this field, by H for hexagonal or R for rhombohedral. The number indicates the total number of silicon carbide layers in the (0001) direction which repeats one each of the unit crystal cells.

The three most generally occurring types are the 6H, often called (II modification), the 4H (III modification), both dihexagonal-dipyramidal and the 15R, rhombohedral, ditrigonal-pyramidal. These occur together in varying percentages as a result of variations in growth conditions. These include changes in temperature, pressure, impurities present in the charge or excess of either silicon or carbon. Nitrogen is the major impurity and therefore pure material must be synthesized in an ambient inert gas (argon is used). Doping, with nitrogen, aluminum, boron or phosphorus is controlled. Twinning is common; there are three kinds, all on the basal plane and on three edges.

The beta form is body-centered cubic and very dense. Its yellow color is not due to any impurity, but to a shift of the fundamental absorption edge into the visible at ~ 4400 Å. Following are the four best lattice constant values:

4.350 - very pure powder 4.352 - colourless crystal

4.349 - powder

4.348 - crystal

[1]

The alpha form has a tetrahedral lattice with silicon in the center and carbon at the four corners like silicon dioxide.

6H	a _o (kX)	د _و (ا	kX)	4H	a _o (kX)	c _o (kX)	
	3.075	15.0	088		3.0739	10.061	
	3.073	15.0	068		3.076	10.046	
	3.073	15.0	079		3.073	10.053	
	3.076	15.0	07 0		3.08	10.09	
	3.08	15.3	17		3.09	10.09	
	3.09	15.	17				[1]
	15R	Н	a _o (kX)	R	a _o (kX)	α	
			3.076		12.78	13°55'	
			3.073		12.703	130541	
			3.073		12.691	13054.5'	
			3.09		12.78	13055'	
			3.08		12.775	13°52'	[1]

Ramsdell³ in 1952 reported five hexagonal modifications including 8H, with an eight layer hexagonal unit cell,

$$a_0 = 3.073 \text{ kX}$$
 $c_0 = 20.106 \text{ kX}$

In 1953⁴ he reported 19H with a structure showing the zigzag sequence 22232323,

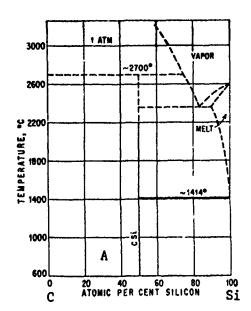
$$a_0 = 3.073$$
 $c_0 = 47.75 \text{ kX}.$

In 1964 Chang-Lin Kuo described two new polytypes:

417R,
$$a_0 = 3.0806 \text{Å}$$
 $c_0 = 1050.7 \text{Å}$
453R, $a_0 = 3.0806 \text{Å}$ $c_0 = 1141.4 \text{Å}$

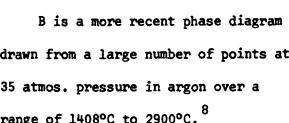
These last references indicate the complexity and the continued interest in the crystallography of silicon carbide.

The commercial material ranges in color from yellow thru green and blue to

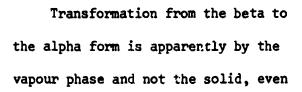


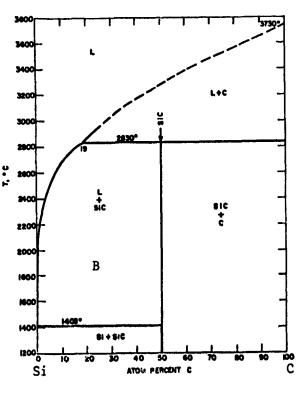
black. Silicon carbide is uniaxial positive and pleochroic.6

The phase diagram A is drawn from four points taken at 1 atm. pressure.



B is a more recent phase diagram : drawn from a large number of points at 35 atmos. pressure in argon over a range of 1408°C to 2900°C.





at temperatures up to the melting point. The beta form is grown below 2000°C from the vapour and forms plates. 9 Silicon carbide crystals are not piezoelectric.

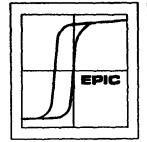
GMELINS HANDBUCH DER ANORGANISCHEN CHEMIE; achte völlig neu bearbeitete Auflage. Silicium. Teil B. Weinheim, Verlag Chemie, GMBH, 1959.

KNIPPENBERG, W.F. Growth Phenomena in Silicon Carbide. PHILIPS RES. REPTS., v. 18, no. 3, June 1963. p. 161-274. [Acces. no. 17402]

RAMSDELL, L.S. and J.A. KOHN. Developments in Silicon Carbide Research. ACTA CRYST., v. 5, 1952. p. 215-224.

RAMSDELL, L.S. and R.S. MITCHELL. A New Hexagonal Polymorph of Silicon Carbide, 19H. AMERICAN MINERALOGIST, v. 38, 1953. p. 56-59.

- CHANG-LIN KUO. Two New Polytypes of Silicon Carbide 417R and 453R. WU LI HSUEH PAO, v. 20, 1964. p. 444-446.
- DANA, J.D. The System of Mineralogy of James Dwight Dana and Edward Salisbury Dana. Yale University, 1837-1892. v.l, 7th ed. Entirely rewritten and greatly enlarged by PALACHE, C., et al. N.Y., Wiley, 1944. p. 123.
- NOWOTNY, H., et al. MONATSH CHEM., v. 85, 1954, p. 255-272 as cited in CONSTITUTION OF BINARY ALLOYS, by HANSEN, M., N.Y., Mc Graw-Hill, 1958.
- SCACE, R.I. and G.A. SLACK. The Silicon Carbide and Germanium Carbide Phase Diagrams. In: CONF. ON SILICON CARBIDE, BOSTON, 1959. SILICON CARBIDE, A HIGH TEMPERATURE SEMICONDUCTOR, PROC., ed. by O'CONNOR, J.R. and J. SMILTENS. N.Y., Symposium Pub. Div., Pergamon Press, 1960. p. 366-371. [Acc. no. 17418]
- GORIN, S.N. and A.A. PLETYUSHKIN. Structural Characteristics of Cubic Silicon Carbide Crystals Grown from the Vapor Phase. IZV. AKAD. NAUK SSSR., SER. FIZ., v. 28, 1964. p. 1310-1315.



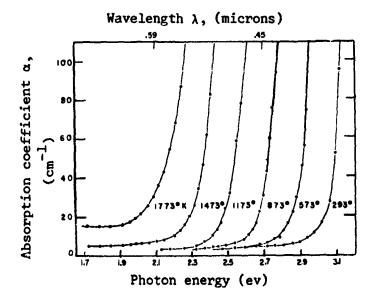


PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

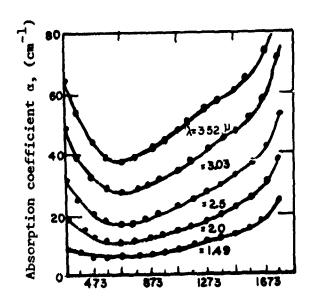
SILICON CARBIDE

ABSORPTION

Absorption as a function of wavelength for pure, colourless, single crystal, alpha-silicon carbide (6H), at several temperatures.



[Ref. 4434]

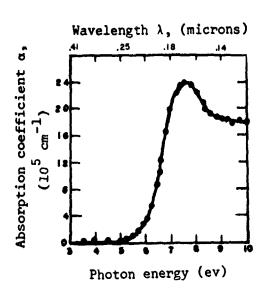


[Ref. 4434]

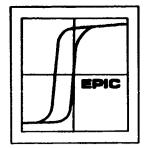
Intrinsic absorption as a function of wavelength at 300°K, for single crystal, alpha-silicon carbide (6H), cut parallel to c-axis. Above 5 ev points are calculated from reflectivity data.

Temperature (°K)

Absorption as a function of temperature for pure, colourless, single crystal alpha-silicon carbide (6H), at several wavelengths.



[Ref. 17419]



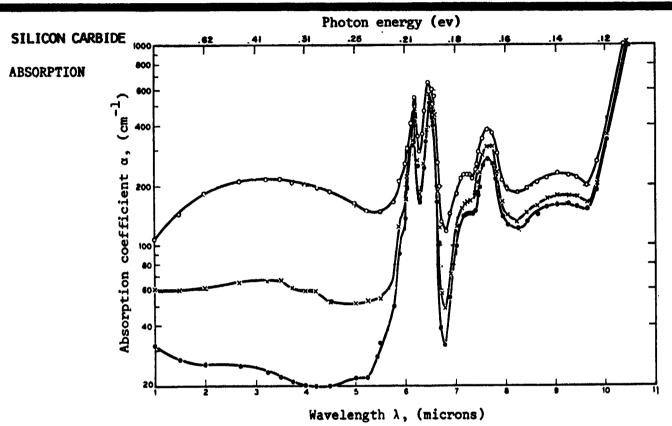
E LECTRONIC

P ROPERTIES

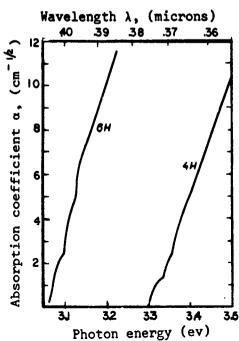
I NFORMATION

C ENTER

PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CAL; FORNIA



Absorption as a function of wavelength for single crystal, n-type, alpha-silicon carbide, (6H). Curves calculated from transmission data, at 300°K. An N-doped inhomogeneous green crystal is used, $\rho \sim 1$ ohm-cm. The 3 curves indicate varying carrier concentrations. [Ref. 628]

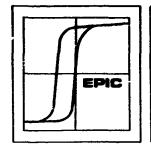


Absorption as a function of wavelength for single crystal, alpha-silicon carbide at $77^{\circ}K$, $E_{\rm i}$ c-axis.

The two curves indicate the absorption edge difference between the (6H) and (4H) polytype.

[Ref. 8792]

AIR PORCE NATERIALS LABORATORY
RESEARCH OF SYSTEMS COMMAND





PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

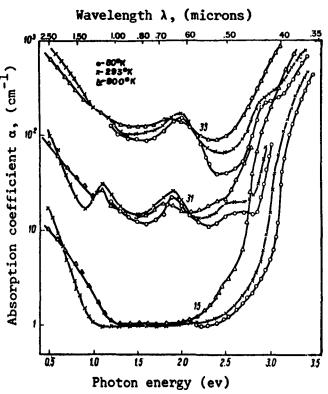
ABSORPTION

Absorption as a function of wavelength for single crystal, n-type, alphasilicon carbide, nitrogen-doped.

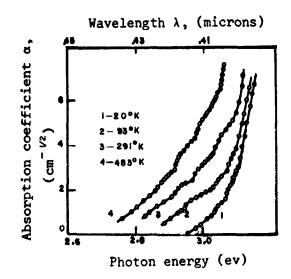
carrier concentration

$$15 - 4.5 \times 10^{17}/cc$$

$$31 - 40 \times 10^{17}$$

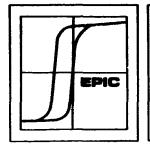


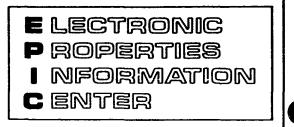
[Ref. 16397]



Absorption as a function of wavelength for single crystal, alpha-silicon carbide, (6H), at four temperatures.

[Ref. 13404]





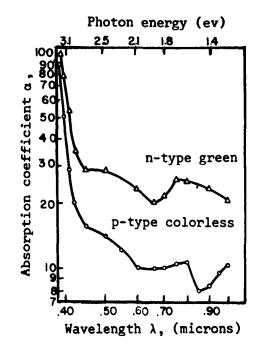
PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

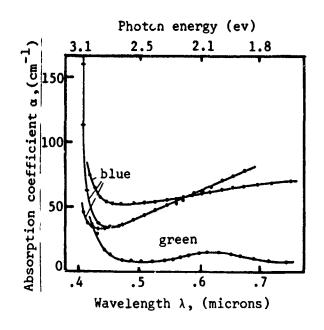
ABSORPTION

Absorption as a function of wavelength for single crystal, alpha silicon carbide at 300°K.

p-type colorless, $\rho = 10^4$ ohm-cm, $n \sim 10^{14}/cc$. n-type green, $\rho = 10^2$ ohm-cm.



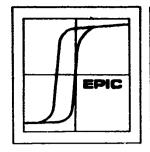
[Ref. 2863]



Absorption as a function of wavelength for single crystal alpha silicon carbide, (6H), at 300°K. Blue crystals are ptype, aluminum-doped, ρ = 50 ohm-cm, green is n-type, ρ = 3 ohm-cm.

[Ref. 2514]

AIR PORCE NATERIALS LABORATORY
RESEARCH OF SYSTEMS COMMAND

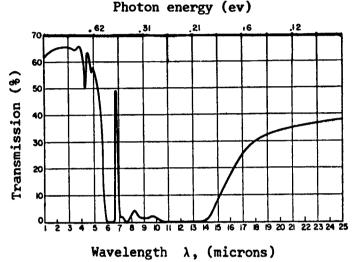


E LECTRONIC P ROPERTIES I NFORMATION C ENTER

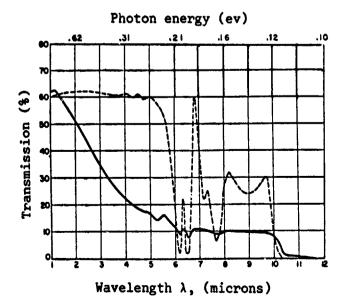
PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER • HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

ABSORPTION



[Ref. 17415]



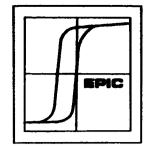
Infra-red transmission of clear and light green, nitrogen-doped, single crystal, n-type, alpha silicon carbide, (6H), at 300°K.

---- clear, 0.07 mm thick

green, 0.007 mm thick.

[Ref. 17415]

AIR PORCE NATERIALS LABORATORY RESEARCH OF STATEMENT OF THE STATEMENT OF T





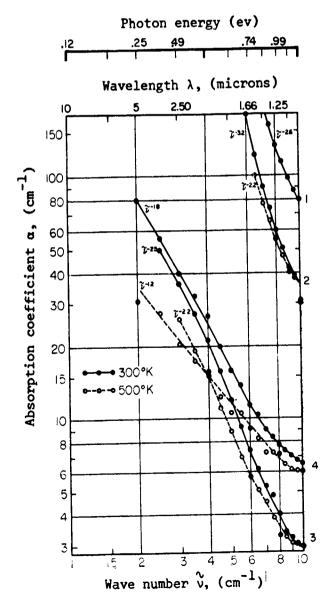
C ENTER

PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

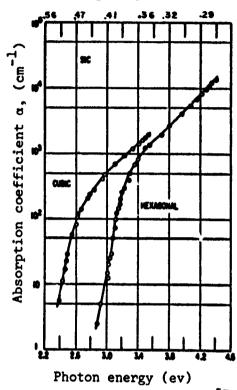
SILICON CARBIDE

ABSORPTION

Absorption as a function of wavelength for single crystal, alpha and beta-silicon carbide, at 300°K. The alpha form is (6H), polytype.



Wavelength λ , (microns)



[Ref. 787]

Absorption as a function of wavelength for single crystal, n-type, silicon carbide. 1, 2, and 3 are alpha-form, commercial grade.

$$1 - n = 6.7 \times 10^{18}/\text{cc}$$

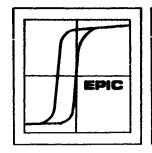
$$2 - n = 2.5 \times 10^{16}$$

$$3 - n = 9.5 \times 10^{16}$$

4 is pure beta-form, $n = 1.3 \times 10^{17}/cc$.

The wave number power on the curves shows the absorption slope.

[Ref. 17233]



E LECTRONIC

P ROPERTIES

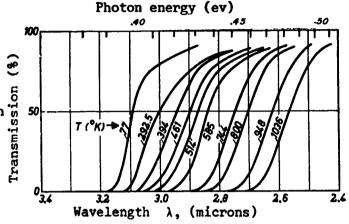
I NFORMATION

PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER • HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

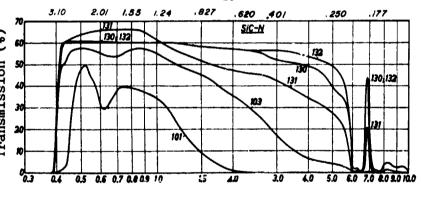
ABSORPTION

Transmission as a function of wavelength of for single crystal, n-type, alpha-silicon of carbide at various temperatures.



[Ref. 3607]

Transmission as a function of wavelength for single crystal, n-type, alpha-silicon carbide at 300°K. 132 is pure, argon grown. The other samples show increasing amounts of the nitrogen dopant.

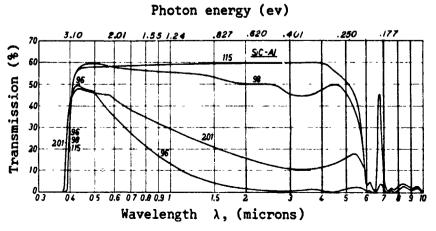


Photon energy (ev)

Wavelength λ , (microns)

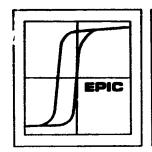
[Ref. 3607]

Transmission as a function of wavelength for single crystal, p-type, alpha-silicon carbide at 300°K. 115 is pure, argon grown, 98, 201, and 96 show increasing amounts of the aluminum dopant.



[Ref. 3607]

AIR PORCE NATURALS LABORATORY
RESEARCH AND TECHNOLOGY DIVISION

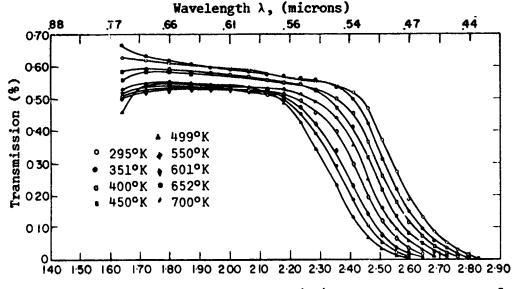


E LECTRONIC
P ROPERTIES
I NFORMATION
C ENTER

PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

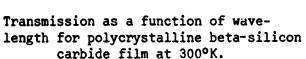
ABSORPTION

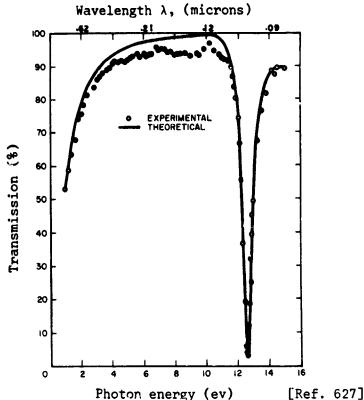


Photon energy (ev)

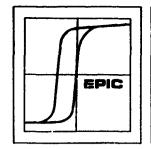
[Ref. 18056]

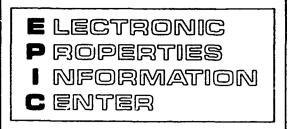
Transmission as a function of wavelength for single crystal, n-type, beta-silicon carbide at temperatures from 295-700°K. n $\sim 10^{17}/cc$, $\rho_{500K} \sim 0.1$ ohm-cm.











PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

CROSS SECTIONS

Symbol Value

Sample

Test Method

Temperature Ref.

 $\sigma \sim 10^{-23} \text{ cm}^2$

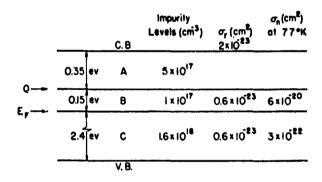
Al-doped, single crystal, α-silicon carbide p-n junction

electroluminescence

300°K

368

calculated for intrinsic recombination



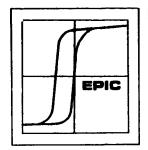
Impurity level densities, and recombination crosssections for both radiative and non-radiative transitions at 77°K. derived from a study of the electroluminescence of single crystal, alpha-silicon carbide p-n junctions between 77-830°K, 0.04 to 2.5 microns and 100 microamps/cm² to 3 amps/cm².

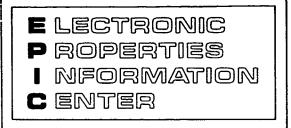
Q - the Quasi-Fermi level

E_r - the Fermi level

σ^r_r - average cross-section for holeelectron radiative recombination

on - average cross-section for holeelectron non-radiative recombination





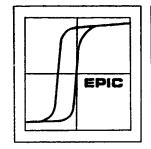
PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

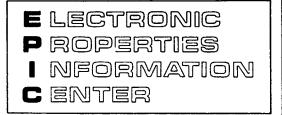
SILICON CARBIDE

DEBYE TEMPERATURE θ

Value	Sample	Test Method	Temperature	Ref.
1200°K	high purity, single crystal	calc. from thermal conductivity meas.	>300°K	17406
913 ° K		calc. from specific heat measurement	53-73°K	*Gmelin
949 ° K			73.5°K	
1192 ° K			700-900°K	

^{*} GMELINS HANDBUCH DER ANORGANISCHEN CHEMIE; achte völlig neu bearbeite Auflage. Silicium. Teil B. Weinheim, Verlag Chemie, GMBH, 1959. p. 820.





PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

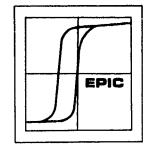
SILICON CARBIDE

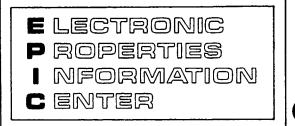
DIELECTRIC CONSTANT

Symbol	Value		Sample	Test Method	Temperature	Ref.
ε ₀	10.62	27**	single crystal, pure	l ke	20°K	3410
	10.38	17	£	5		ŧ
	10.26	5 3		32		
	10.23	3		100		
	10.99	49		1	77°K	
	10.69	29		1 5		+
	10.42	14		32		
	10.34	8		100		3410
εο	10.0 ±	0.1	single crystal, α ,	IR optical, 1-25μ	300°K	628
			$\rho > 1$ ohm-cm	λ > 12.6μ		
€	6.7			λ < 12.6μ		628
ε	6.7		epitaxial single crystal film	IR optical measurement at 1-15μ	nt	627
	50-200		powder	up to 30 kc		*Gmelin

^{**} This column shows tan $\delta \times 10^{-3}$ values for the sample under the given conditions.

^{*}GMELINS HANDBUCH DER ANORGANISCHEN CHEMIE; achte völlig neu bearbeitete Auflage. Silicium. Teil B. Weinheim, Verlag Chemie, GMBH, 1959. p. 829.





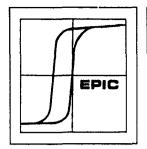
PREPARED BY ELECTPONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

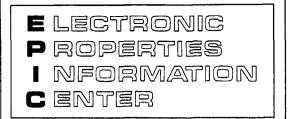
SILICON CARBIDE

EFFECTIVE MASS

Symbol	Value	Sample	Test Method	Temperature	Ref.
m≉	2.5 (calc.)	<pre>single crystal, p-, n-type, n = 8 x 10¹⁹/cc</pre>	elec. conductivit and Hall measurem	•	2419
m _n	0.6	<pre>single crystal, n-, and p- type</pre>	u	300°K	3608
m P	1.2-2	single crystal, α, N-, B- or Al-doped	II	11	3608
n/ * P	∿1.0	single crystal, p-type, α ,	mobility meas. from Hall cond. data	77-1500°K	8783
mn	0.72	Al- and B-doped			
m _D	3.5				8783

^{* -} density of states Effective Mass





PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

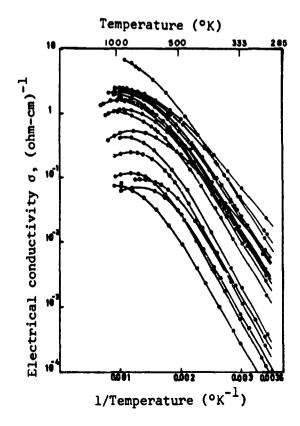
ELECTRICAL CONDUCTIVITY

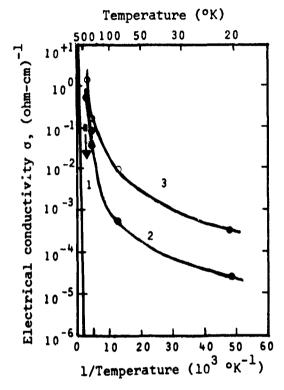
Electrical conductivity as a function of temperature for single crystal, alphasilicon carbide.

1 - n-type, N-doped,
$$n \sim 10^{17}/cc.$$
, (6H)

2 - n-type, N-doped,
$$n \sim 10^{19}/cc.$$
, (6H)

3 - n-type, Al-doped,
$$n = 4 \times 10^{19}/cc.$$
, (6H) and (15R).



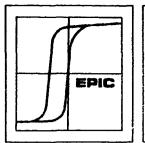


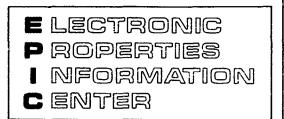
[Ref. 17725]

Electrical conductivity as a function of temperature for p-type, single crystal, alpha-silicon carbide.

$$n = 2 \times 10^{17} \text{ to } > 10^{20}/\text{cc.}$$

[Ref. 18406]



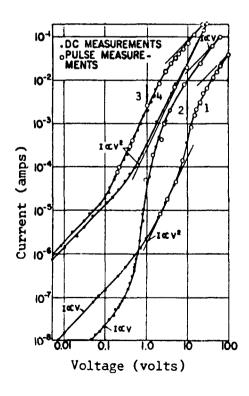


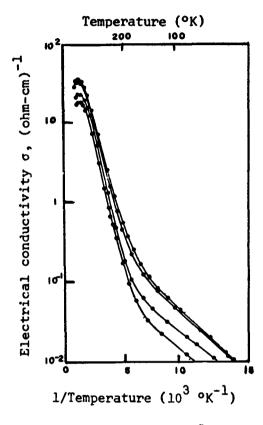
PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

ELECTRICAL CONDUCTIVITY

Electrical conductivity as a function of temperature for single crystal, alphasilicon carbide, black crystals, with varied cross-section and carrier concentration.



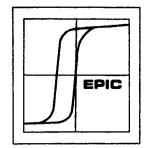


[Ref. 6080]

Current voltage curves of single crystal, alpha-silicon carbide at 300°K.

				p ohm-cm
1	-	p-type,	B-doped,	670,000
2	-	11	11	53,000
3	-	*1	11	11,000
4	-	11	undoped	5,300

[Ref. 5908]



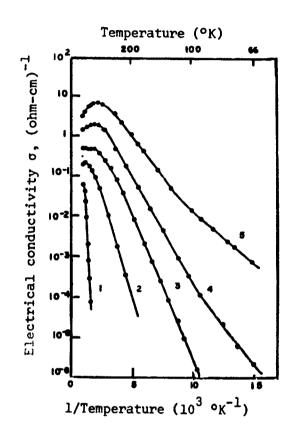


PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

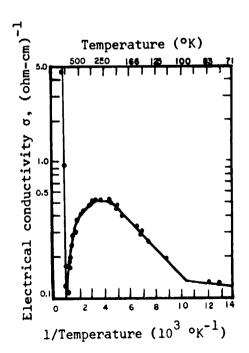
SILICON CARBIDE

ELECTRICAL CONDUCTIVITY

Electrical conductivity as a function of temperature for single crystal, alpha and beta-silicon carbide, pale yellow to green. Samples from 1 to 5 have varied cross-section and show increase in impurity by colour change from pale yellow to dark green.

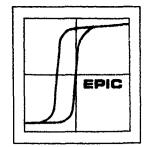


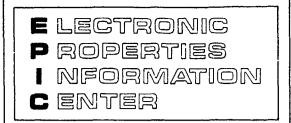
[Ref. 6080]



Electrical conductivity as a function of temperature for n-type, fairly pure, beta-silicon carbide.

[Ref. 17725]

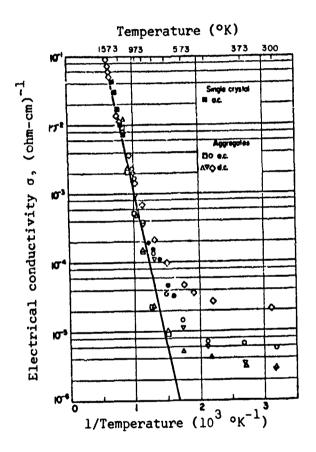




PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

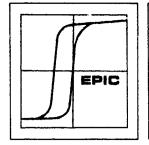
ELECTRICAL CONDUCTIVITY



Electrical conductivity as a function of temperature for beta-silicon carbide, both single crystal, and polycrystalline.

[Ref. 17416]

AIR FORCE MATERIALS LABORATORY RESEARCH AND TECHNOLOGY DIVISION



E LECTRONIC
P ROPERTIES
I NFORMATION
C ENTER

PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

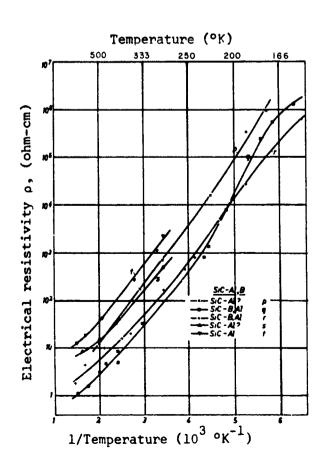
ELECTRICAL RESISTIVITY

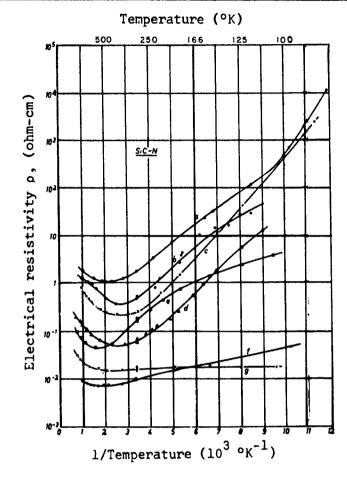
Resistivity as a function of temperature for n-type, single crystal, alpha-silicon carbide.

a) is grown in pure argon, no dopants, the other curves show increasing amounts of nitrogen,

n from 10^{17} to 5 x $10^{19}/cc$.

d) and e) also have aluminum and boron impurities.





[Ref. 3608]

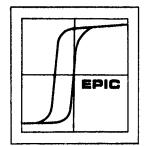
Resistivity as a function of temperature for single crystal, alpha-silicon carbide.

- p) and s) are grown in pure argon
- q) and r) in argon and boron
- t) in argon + aluminum.

n varies from 10^{17} to $10^{18}/cc$.

[Ref. 3608]

AIR FORCE MATERIALS LABORATORY
RESEARCH AND TECHNOLOGY DIVISION

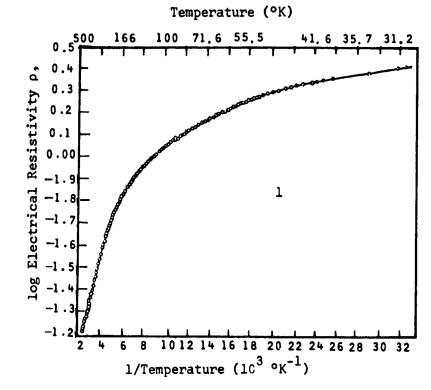


PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

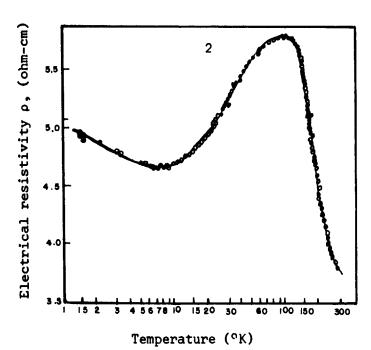
ELECTRICAL RESISTIVITY

1) $n = 6.4 \times 10^{16}/cc$.



Resistivity as a function of temperature for n-type, single crystal, alpha-silicon carbide, fairly pure, slightly N-doped.

[Ref. 17973]

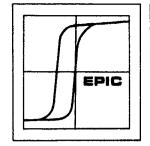


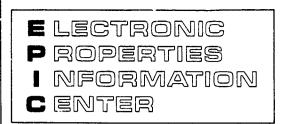
2) $n = 3 \times 10^{17}/cc$.

[Ref. 17973]

AIR FORCE MATERIALS LABORATORY RESEARCH AND TECHNOLOGY DIVISION

RESEARCH AND TECHNOLOGY DIVISION AIR FORCE SYSTEMS COMMAND



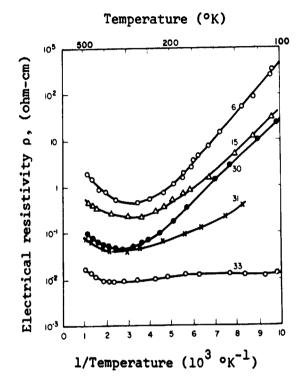


PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

ELECTRICAL RESISTIVITY

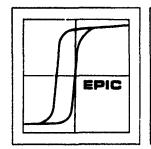
Resistivity as a function of temperature for n-type, single crystal, nitrogendoped, alpha-silicon carbide, (6H).

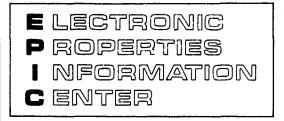


[Ref. 16397]

Approximate Resistivity (ohm-cm)	Dopant in Single crystal, β-SiC (ppm)	Crystal Type	Color	Temperature
0.5	60 B	p	green	300°K
10 ⁻¹	200 Al	P	blue-green	ı
10	< 10 Al	P	yellow-green	
10 ³	no Al detected	n	light yellow	
103	no B detected	n	light yellow	•

[Ref. 17725]





PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

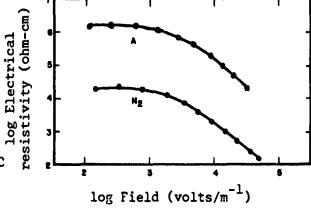
SILICON CARBIDE

ELECTRICAL RESISTIVITY

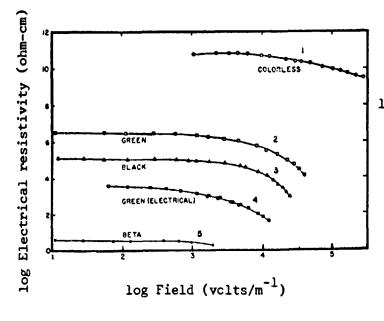
Log electrical resistivity as a function of electric field for pure, alpha-silicon carbide grains.

A - grains heated 2.5 hours at 1950°C in argon

N₂ - grains heated 2.5 hours at 1950°C in nitrogen gas.



[Ref. 17403]

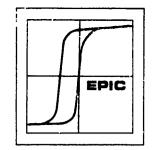


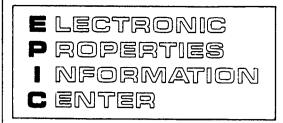
Log electrical resistivity as a function of field for commercial silicon carbide grains.

5 is beta-silicon carbide grain.

[Ref. 17403]







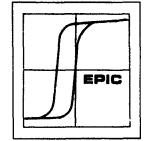
PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

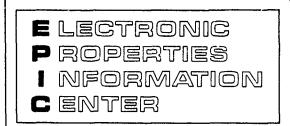
SILICON CARBIDE

ELECTROACOUSTIC PROPERTIES

Pure, and nitrogen-doped, n-type, single crystal, α -SiC, (6H), at 6°K. $n = 10^{15}$ to $10^{17}/cc$. Phonon energies in ev for various k vectors from luminescent

data.				
Phonon branc	$k_c = 2$	$2\pi/c k_c = 4\pi/c$	$k_c = 6\pi/c$	
TA	.0335	.0392	.044	
	.0363	.0403	.0463	
LA	.0506	.0670	.077	
	.0535	.069		
то	.0978	.0956	.0947	
LO	.107	.1055	.1042	[Ref. 6505]
Phonon branch	Value	Sample	Test Method	Temperature
LO	0.12 ev	n-type, N-doped, α -,	reflectivity data	300°K
TO	0.095			
				[Ref. 3607]





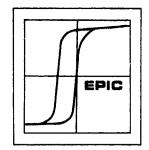
PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

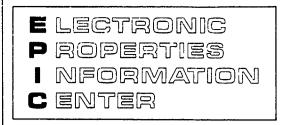
SILICON CARBIDE

ELECTROACOUSTIC PROPERTIES

Phonon energies for single crystal, n-type, nitrogen-doped, α and β -silicon carbide at $6^{\circ}K$.

		α		
Phonon branch	β	15R	6Н	Ref.
TA	.0463 ev	.0463 ev	.0463 ev	17405
LA	.0794	.0782	.077	
TO	.0944	.0946	.0947	
LO	.1028	.1037	.1042	17405
TA		.0344	.0335	13156
TA		.0350	.0363	
TA		.0393	.0392	- 1
TA		.0397	.0403	
TA		.0432	.044	
TA		.0463	.0463	
LA		.0513	.0506	
LA		.0519	.0535	
LA		.0692	.067	
LA		.0702	.0690	
LA		.0782	.077	
TO		.0946	.0947	
TO		.0953	.0956	
TO		.0957	.0978	
TO		.0971		
LO		.1037	.1042	
LO		.1063	.1055	•
LO		.1069	.107	13156

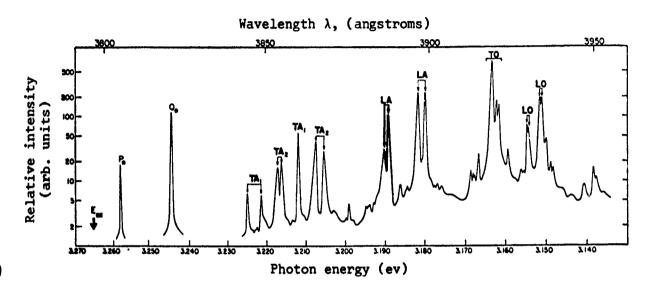




PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

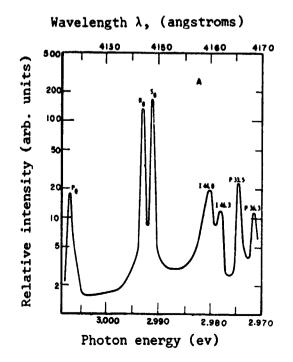
SILICON CARBIDE

ELECTROACOUSTIC PROPERTIES



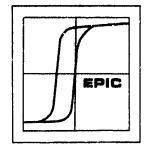
[Ref. 17974]

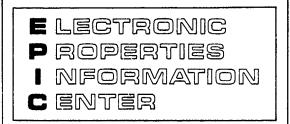
Luminescence spectrum at 6°K of four-particle nitrogen-exciton complexes in single crystal, alpha-silicon carbide, (4H). P_0 and Q_0 are zero-phonon lines.



Luminescence spectrum of pure, single crystal, alpha-silicon carbide, (6H), at 6° K, n = 10^{15} /cc. The two additional intrinsic lines (I 44.0 & I 46.3) are due to recombination of free excitons.

[Ref. 6505]



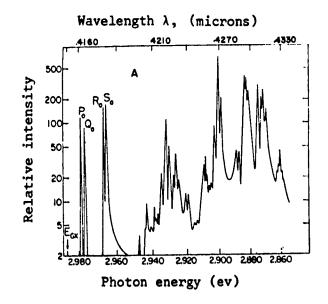


PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

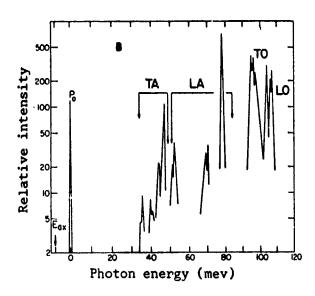
SILICON CARBIDE

ELECTROACOUSTIC PROPERTIES

Complete four-particle nitrogen-exciton luminescence spectrum at 6°K, for n-type, single crystal, alpha-silicon carbide, (15R). The four phonon-free lines are marked. Exciton energy gap = 2.98 ev.



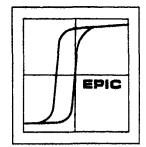
[Ref. 13156]

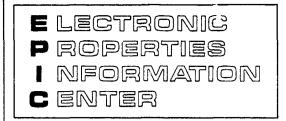


One of the four series of luminescence lines shown in the P series of A. The zero of energy has been placed at P₀, so that the energies of emitted phonons may be read directly.

[Ref. 13156]

AIR FORCE MATERIALS LABORATORY RESEARCH AND TECHNOLOGY DIVISION

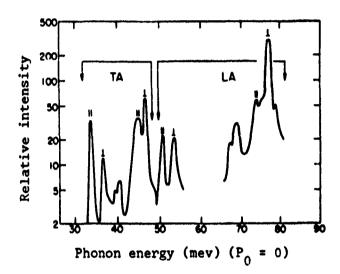




PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

ELECTROACOUSTIC PROPERTIES



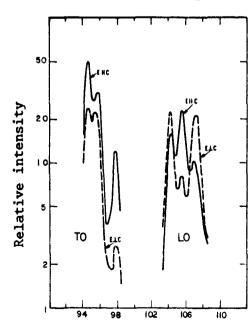
The phonon energy in ev represents the displacement of the energy from that of the phonon free line P_0 . Polarization is parallel or normal to the c-axis.

Optical phonon lines _____

Samples are pure, single crystal, alphasilicon carbide, (6H), at 6° K. $n = 10^{15}/cc$.

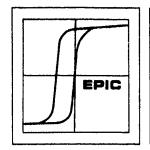
____ Acoustic phonon lines

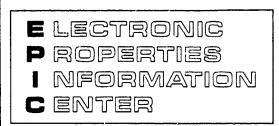
[Ref. 6505]



Phonon energy (mev) $(P_0 = 0)$

[Ref. 6505]



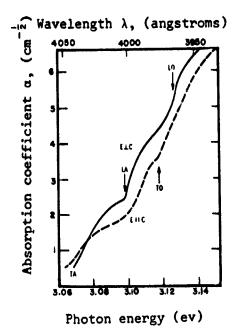


PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

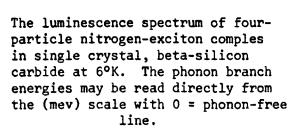
SILICON CARBIDE

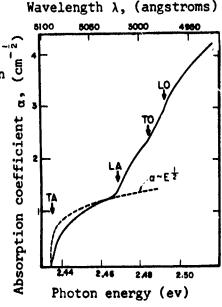
ELECTROACOUSTIC PROPERTIES

Absorption edge of single crystal, n-type, beta-silicon carbide, at 4.2°K. Arrows indicate possible phonon emission. Exciton energy gap is 2.39 ev.



alpha-silicon carb

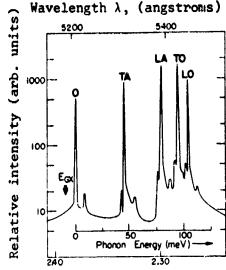




[Ref. 17405]

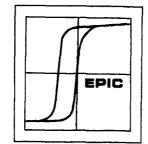
Absorption edges of pure, n-type, single crystal, alpha-silicon carbide, (6H), in polarized light at 77°K.

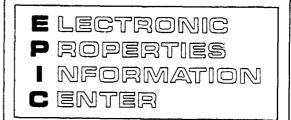
n = 10¹⁵/cc.



Photon energy (ev)

[Ref. 17405]





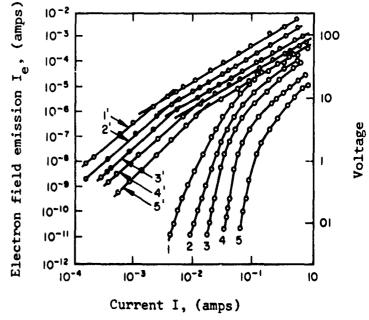
PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

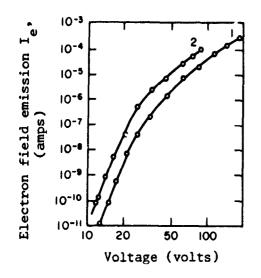
ELECTRON-FIELD EMISSION

Electron emission as a function of field in single crystal, alphasilicon carbide p-n junction, formed by boron diffusion, (curves 1 to 5).

Voltage-current characteristics for the same sample are shown by curves 1' to 5'.



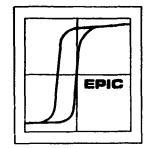
[Ref. 17414]

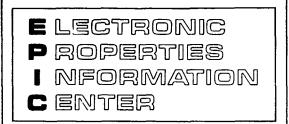


Electron emission as a function of field for single crystal, alpha-silicon carbide p-n junction at two temperatures.

1 - 293°K

2 ~ 673°K

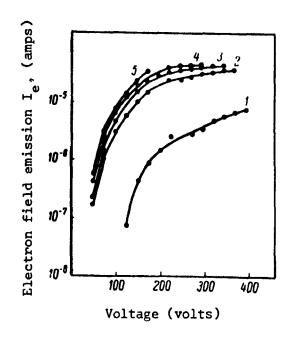




PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

ELECTRON-FIELD EMISSION



Electron emission as a function of field for single crystal, alpha-silicon carbide p-n junction.

1 - 293°K

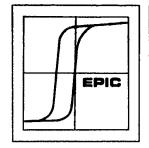
2 - 473°K

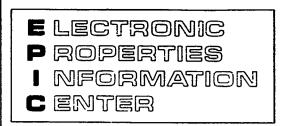
3 - 553°K

4 - 663°K

5 - 773°K

[Ref. 14663]



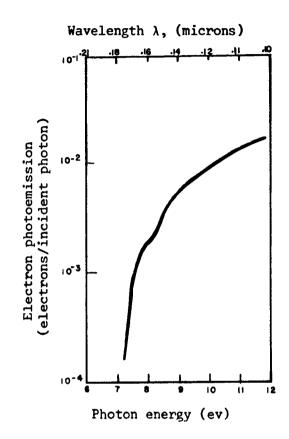


PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER • HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

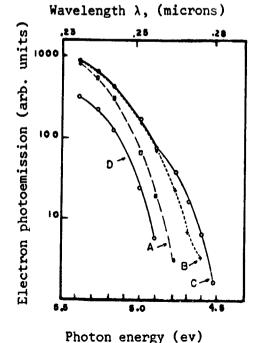
SILICON CARBIDE

ELECTRON PHOTOEMISSION

Electron photoemission as a function of wavelength for single crystal, (yellow color), beta-silicon carbide at 300°K.



[Ref. 787]



Electron photoemission as a function of wavelength in n-type, single crystal, (green color), alpha-silicon carbide at 300° K. $\rho = 0.6$ ohm-cm. The surfaces were variously treated by ion bombardment alone, and with annealing, as well as exposure to oxygen.

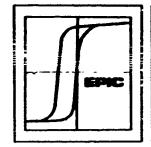
A - ion bombardment

B - ion bombardment + 1000°C anneal at 2 hrs.

C - ion bombardment + 1000°C anneal at 20 hrs.

D - oxygen exposure

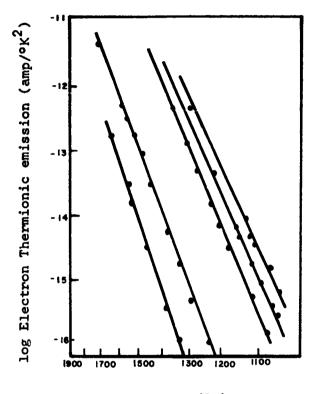
[Ref. 7134]



PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER • HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

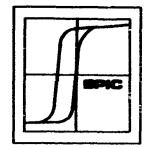
ELECTRON THERMIONIC-EMISSION

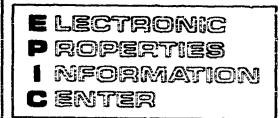


Temperature (°K)

Electron thermionic emission in single crystal, alpha-silicon carbide, (commercial grade). Data at increasing temperatures are not reversible and fresh samples are required for each run.

[Ref. 7152]





PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

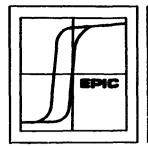
. SILICON CARBIDE

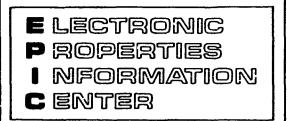
EMERGY BANDS

-	Value (ev/°K)	Sample (single crystal) n-type, α-, n = 10 ¹⁷ /cc	Test Measurement	Temperature Ref. 295-700°K 18056
		ρ ~ 0.1 ohm-cm	.885 to 428µ	
		n-type, α-, (6H), N-doped n ∿ 10 ¹⁷ /cc nd (001) oriented	resistivity, Hall meas. at 10 oe, al optical absorp. at .35 to 2.5µ	.so
dEg/dT	-3.3 x 10 ⁻⁴	α-,	transmission and photoelectric	>300°K 17417
dEg/dT	-3 to -6 x 10 ⁻⁴	n-, and p-type, α	photoconductivity $\lambda = 0.6$ to 5.5 ev	
	- 3 to -4 x 10 ⁻⁴	n	p-n junction photo effect) 11
dEg/dT	-3.3 to 3.4 x 10 ⁻⁴	n-type, α-, (6H), (001) oriented,	absorption	100°K, 293°K, 4596 & 500°K
dI/dT	-3.8 ± 0.6 x 10 ⁻⁴	ρ _{300K} = 7.5 ohm-cm	photoconductivity	4596
E _{dc}	11.5 e v	p-type, α-, (6H). pure, (Cl in argon, Al, 0 ₂ , N im		300-1300°K 8783
E _{dv}	13	2	-	

39

 E_{dc} = deformation potential of conduction band E_{dv} = deformation potential of valence band



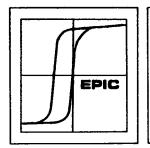


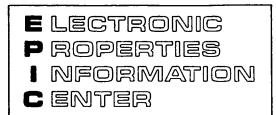
PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

EMERGY BANDS

Symbol	Value (ev/°K)	Sample (single crystal)	Test Measurement	Temperature	Ref.
E ₁ -E ₂	.0048 e▼	spin orbit splitting of walence band	photolum.	6 ° K	6505
k0	0.09 ev	indirect transition in an α-, (6H) sample	transmission and photoelectric	77-717 ° K	17417

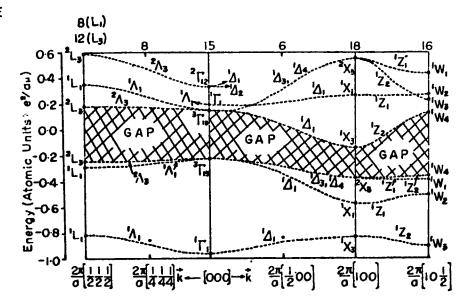




PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER • HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

ENERGY BANDS



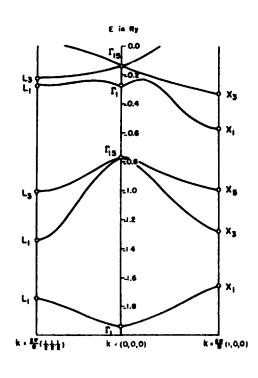
Energy band structure of beta silicon carbide crystal obtained by OPW-calculation.

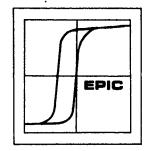
The numbers at the top side of the figure represent those of the OPW-terms involved.

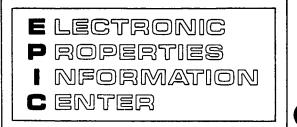
[Ref. 2348]

Ry	ev
.2	2.7
.4	5.4
.6	8.1
.8	10.8
1.	13.5
1.2	16.2
1.4	18.9
1.6	21.6
1.8	24.3

Band structure of silicon carbide. The computed values for the electronic states at the symmetry points Γ , L, and X are indicated by circles. The lines indicate the expected band profiles in the directions Λ and Λ .





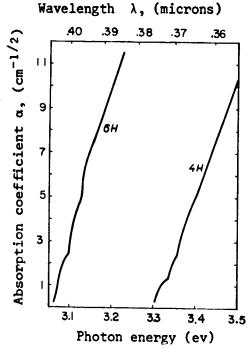


PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER • HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

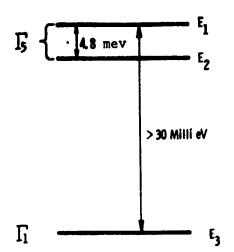
SILICON CARBIDE

ENERGY BANDS

Effect on energy gap of crystalline, polytype. Two samples of alpha silicon carbide, (6H) and (4H), are measured at 77°K, E normal to c-axis.

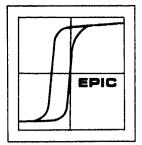


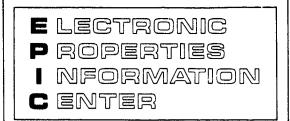
[Ref. 8792]



Observed splittings of the three valence bands in silicon carbide. The small spin-orbit interaction splits \mathbf{E}_1 and \mathbf{E}_2 , and a much larger crystal field interaction splits off \mathbf{E}_3 .

[Ref. 6505]

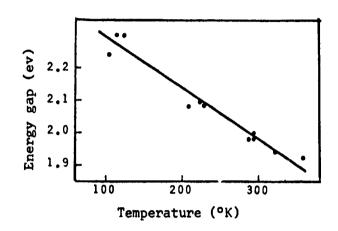




PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

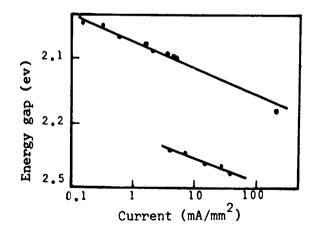
SILICON CARBIDE

ENERGY BANDS



Shift in energy gap as a function of temperature at constant current density for lightly coloured p- and n-type single crystal alpha silicon carbide.

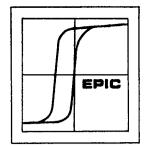
[Ref. 2514]

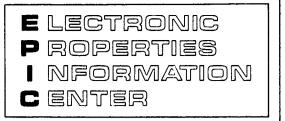


Shift in energy gap as a function of current density for lightly coloured p- and n-type single crystal alpha silicon carbide.

[Ref. 2514]

AIR FORCE MATERIALS LABORATORY
RESEARCH AND SYSTEM OF STREET



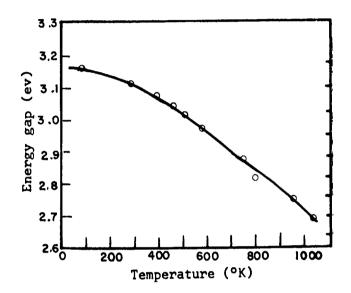


PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

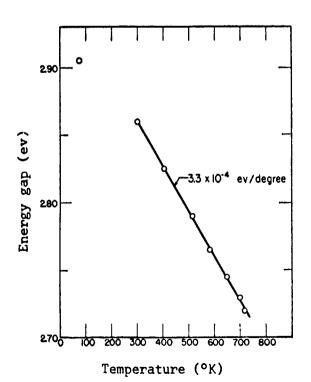
SILICON CARBIDE

ENERGY BANDS

Energy gap as a function of temperature for pure, colorless, alpha-silicon carbide.



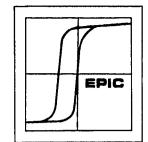
[Ref. 3607]

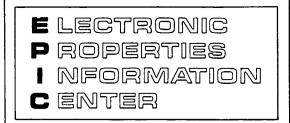


Shift in energy gap as a function of temperature for alpha-silicon carbide.

[Ref. 17417]







PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

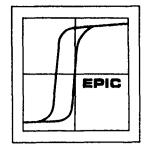
SILICON CARBIDE

ENERGY GAP

Symbol	Value (ev)	Sample (single crystal)	Test Measurement	Temperature	Ref.
Eg	3.12 2.62	α-, (6H) pure, β-,	optical absorp.	300°K	787 787
)	2.86 2.2	α-, (6H) β-,	91 11	"	17419 17419
•	3.1 ± 0.2	n-type, α-,	conductivity, from 950-2000°K	0 ° K	17455
	3 2.8	pure, colorless, α -, yellow, β -,	optical absorp.	300 ° K	3607 3607
	2.986	n-type, N-doped, α-, (15R)	luminescence	6°K	13156
	2.86	α-,	optical	300°K	17417
	2.83 ± 0.1	p-n junction	radiation recomb.	300°K	7333
	2.8	n-type,green,ρ = 10 ² ohm-cm p-type,colorless,ρ = 10 ⁴ ohm-cm	transmission at .221 micron	300 ° K	2863

AIR FORCE MATERIALS LABORATORY

RESEARCH AND TECHNOLOGY DIVISION AIR FORCE SYSTEMS COMMAND



E LECTRONIC
P ROPERTIES
I NFORMATION
C ENTER

PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

ENERGY LEVELS

Explanation of Symbols used in Energy Levels Table

VB - distance above valence band in ev

CB - distance below conduction band in ev

E_C - conduction band edge

 $\mathbf{E}_{\mathbf{V}}$ - valence band edge

 E_r - Fermi level

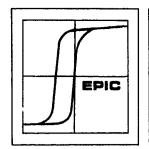
E_n - doncr ionization energy level

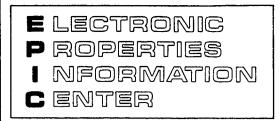
E_A - acceptor ionization energy level

E_{Gx}- exciton energy gap

 $\mathbf{E}_{\mathrm{peak}}$ - photon energy at emission peak

Symbol	Value (ev)	Dopant	Sample (single crystal)	Test Measurement Temperature	Ref.
ED	0.02	N	n-type, β-,	electrical conductivity	17725
ED	0.09	N	n-type, α-,	11	17725
ED		N	n-type, α -, (6H) n = $10^{17}/cc$	Hall & resistivity 100-1000°K meas. at 10 ⁴ oe	16397
	0.086 0.063 0.23		$\rho = .425 \text{ ohm-cm}$ $\rho = 0.2 \text{ ohm-cm}$ $\rho = 0.048 \text{ ohm-cm}$		16397
ED	0.17 0.20	N N	n-type, α-, (6H)	photoluminescence 77°K	11725
	0.23	N	11	11 11	11725





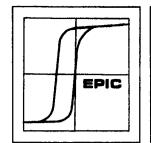
PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

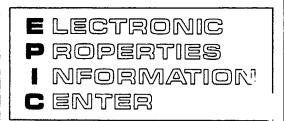
SILICON CARBIDE

ENERGY	LEVELS					
Symbol	Value (ev)	Dopant (s	Sample ingle crystals)	Test Measurements	Temperature	Ref.
EA	0.21	§ Mov	pe, $\rho \sim 0.5$ ohm-cm $n \sim 10^{18}$ /cc	Hall measurements		15717
	0.22	electrons	n ∿ 10 ¹⁸	radia+ion recombina	ition	
	0.23		$n \sim 7 \times 10^{17}$	Hall measurements		
	0.25		n ∿ 7 x 10 ¹⁷	radiation recombina	ition	15717
ED	0.47	fast neutron flux	n-type	resistivity measure	ement 300°K	*
EA	0.24	10 ¹⁷ /cm ²	p-type			
E _F	0.63	p-n	junction, α-,	calculated from	673°K	7333
E _F	0.52	n ∿	8 x 10 ¹⁷ /cc	radiation recomb. & resistivity meas.	300°K	
	.47 to 0	ρ =	10 ⁵ ohm-cm	o resistivity meas	300°K	7333
Symi	bol	Sample, R	esistivity and Colo	Test ur Measurements	Temperatur	e Ref.
El	E ₂	n-type, α-, si	ngle transparent cr	ystals electrical conductivity		6080
1.57		10 ¹³ ohm-cm	pale yellow	1	>1300°K	
0.441		10	pale yellow-g	reen	1100°K	

	Symbol		Sample, Resis	tivity and Colour	Test Measurements	re Ref.	
	E ₁	E ₂		transparent crystals	electrical conductivity	>12000K	6080
	1.57		10 ¹³ ohm-cm	pale yellow	Ť	>1300°K	
	0.441		10	pale yellow-green	ļ	1100°K	
	0.304		10	pale green		625 ° K	
	0.263	0.147	1	dark green		565°K	
•	0.198	0.097	.3	dark green	ļ	480°K	
•			p-type, α-, opaque				
	0.2	0.5	l ohm-cm			∿1000°K	6080

CENTRE D' ETUDE DE L' ENERGIE NUCLEAIRE. MOL, BELGIUM. Solid State Neutron Detector. QR No. 6, Jan. 1 - Mar. 31, 1962. EURAEC-306. Contract No. 079-61-10RDB. 1962.



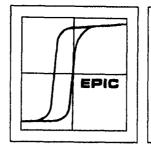


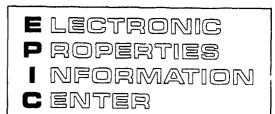
PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

ENERGY LEVELS

Symbol	Valu (ev	ne 7) Dopant	Samp (single c		Test Measurem	ment Temperature	Ref.
$\mathbf{E}_{\mathbf{D}}$	0.14 0.16	N N	n-type, α-,	(15R)	optical absorption		13156
	0.20	N					13156
ED	0.6 0.8 _{CB}	N 3	n-type, α-,	n ~ 10 ¹⁸ /co n ~ 10 ¹⁷ /co	c Hall and conducti	ivity meas.	3608 3608
E _{Gx} -E _{pe}	eak	E _{G×}	E peak	Emission Color	Sample B- & N-doped, α-, powders	Test Measurement Temp	. Ref.
0.95 et		3.263 ev	2.317 ev	green	(4H)		K 17404
0.89 0.91		3.023 3.01	2.137 2.101	yellow	(H)	luminescence	
0.91		2.986	2.101	orange orange-red	(33R) (15R)	at .365 microns	
0.88		2.86	1.983	red	(21R)	1000 11101 0110	
0.91		2.80	(1.89)		(8H)		
0.91		2.390	(1.48)		(3C)		17404
E _A	0.27 _V	B Al	p-type, α-	,	Hall & electrical resistivity	. 300-1300°K	8783 8783
EA	0.275	VB Al	p-type, α-	, n< 10 ¹⁸ /co	: Hall & conductivi measurement	ty	3608
	0.15	N	n-type, α-,		glow curve data	106 ° K	3607
	0.20 0.085	Al N	p-type, α-,		11 £1	233°K	3607
$\mathbf{E}_{\mathbf{D}}$	0.083 0.27	CB N	n-type, α -, p-type, α -,		fluorescence	77°K	3607 3607
A	ν.	В	F -7F-7 - 7				3007



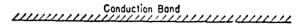


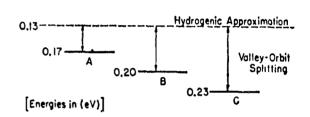
PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

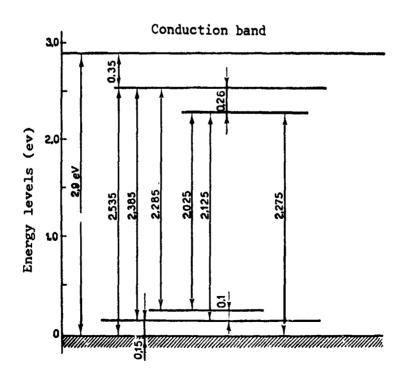
ENERGY LEVELS

Energy level diagram showing the relation of the three nitrogen donor levels [A, B, and C] to that obtained by the hydrogenic approximation. The difference between 0.13 ev and a donor level may be taken as a rough measure of the valley-orbit splitting for that donor.



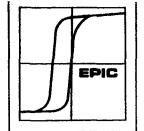


[Ref. 11725]



Valence band

Energy level diagram for electroluminescence in silicon carbide. [Ref. 2899]

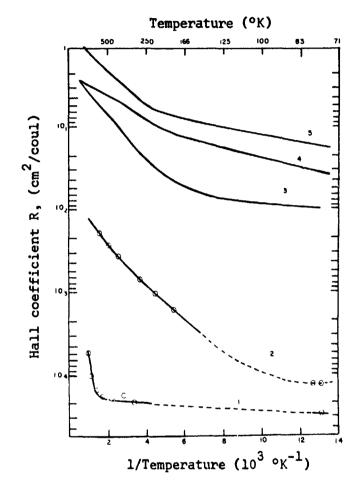




PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

HALL COEFFICIENT



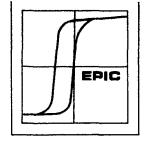
[Ref. 5464]

Hall coefficient as a function of temperature for pure, n-type, single crystal, alpha-silicon carbide.

Dotted lines indicate no experimental points.

AIR FORCE MATERIALS LABORATORY

RESEARCH AND TECHNOLOGY DIVISION AIR FORCE SYSTEMS COMMAND





PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

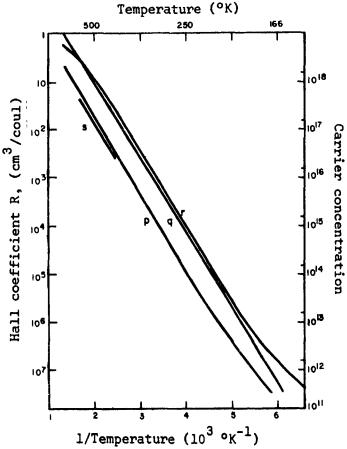
SILICON CARBIDE

HALL COEFFICIENT

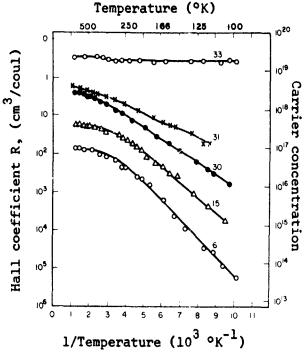
Hall coefficient as a function of temperature for p-type, single crystal, alpha-silicon carbide, (prepared in argon, argon + boron chloride and argon + aluminum chloride).

p - undoped
$$n \sim 10^{17}/cc$$

q - B-doped $n = 2.3 \times 10^{17}$
r - " $n = 9.5 \times 10^{17}$
s - Al-doped $n = 10^{18}$



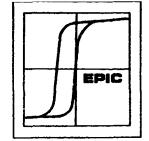
[Ref. 3608]

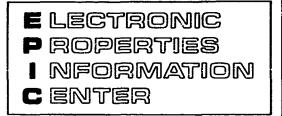


Hall coefficient as a function of temperature for n-type, single crystal, alpha-silicon carbide, (6H).

[Ref. 16397]

AIR FORCE MATERIALS LABORATORY RESEARCH AND TECHNOLOGY DIVISION

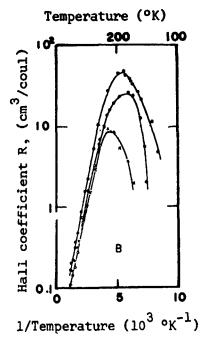




PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

HALL COEFFICIENT



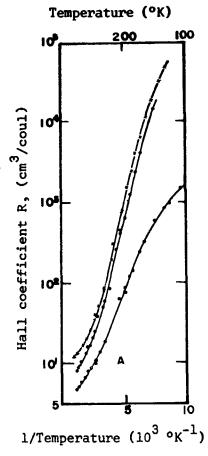
Hall coefficient as a function of temperature for single crystal, alpha-silicon carbide.

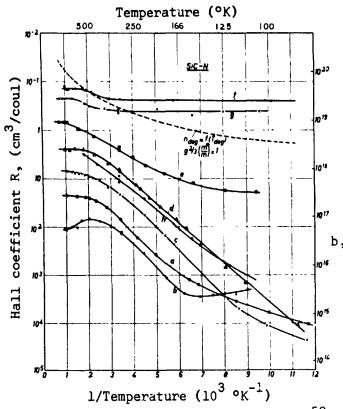
570°K

A - n-type, green

B - p-type, black

 $\rho_{1000K} \sim 0.05$ ohm-cm.





Hall coefficient as a function of temperature for n-type, alpha-silicon carbide, (prepared in argon, argon + nitrogen and for [h] argon, nitrogen and aluminum chloride).

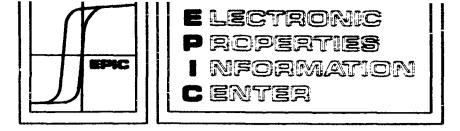
a - undoped
$$n = 3.6 \times 10^{17}/cc.$$
c - N-, Al-, B-, 10^{18}
d - 10^{19}
e - 10^{19}
f - N-doped $n = 5.0 \times 10^{19}$
g - " $n = 3.4 \times 10^{19}$
h - N-, Al-doped $n = 3.0 \times 10^{19}$

The dotted line is a theoretical curve.

[Ref. 3608]

[Ref. 5990]

AIR FORCE NATERIALS LABORATORY
RESEARCH AND TECHNOLOGY DIVISION



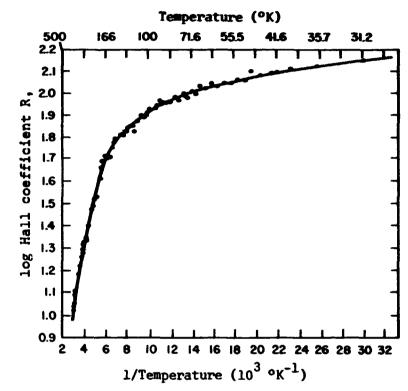
PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER • HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

HALL COEFFICIENT

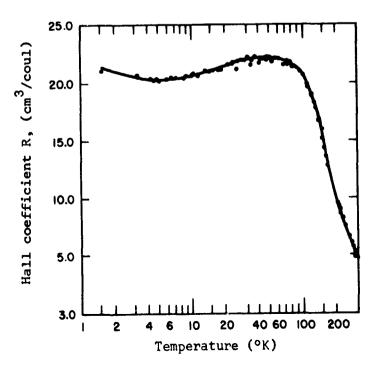
 $n = 5.6 \times 10^{17}/cc.$

 ρ_{290K} = .222 ohm-cm.



Hall coefficient as a function of temperature for n-type, single crystal, alpha-silicon carbide.

[Ref. 17973]

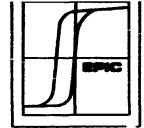


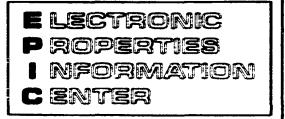
 $n = 1.3 \times 10^{18}/cc.$

 $\rho_{290K} = .0398 \text{ ohm-cm}.$

[Ref. 17973]

AIR PORCE MATHEMALS LABORATORS
RESEARCH AND SECHNOLOSY DIVISION

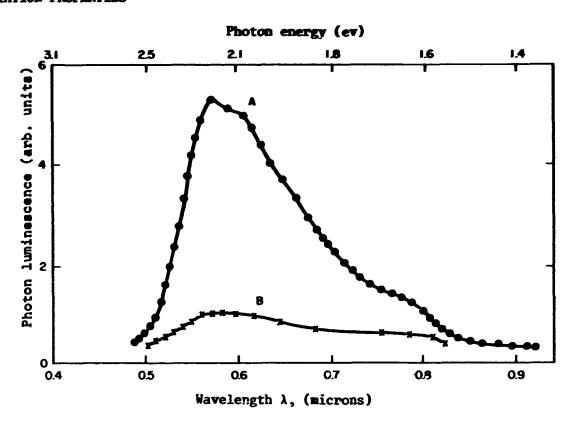




PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

IRRADIATION PROPERTIES



Photoluminescence as a function of wavelength for p-type, single crystal, alpha-silicon carbide at 300° K. The samples were irradiated by 5×10^{11} neutrons/cm² sec.

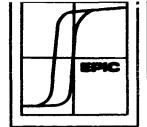
A - characteristic fluorescence of aluminum-doped sample
B - after 10 minutes irradiation

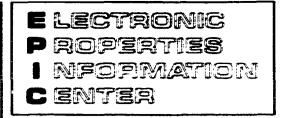
[Ref. 18349]

According to their original quality, silicon carbide diodes are affected by a fast neutron radiation flux of 10¹⁴/cm² to 10¹⁶ or 10¹⁷/cm². *

^{*} CENTRE D' ETUDE DE L' ENERGIE NUCLEAIRE, BRUSSELS. Solid State Detector for Neutrons. QR No. 2, Jan. - Mar. 1961. EURAEC-107. 1961.

AIR PURCE NATIONALS LABORATORY RESEARCH AND TYSTONIS GO DIVISION

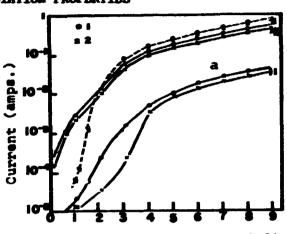


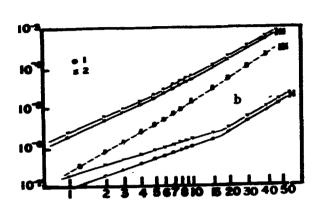


PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

IRRADIATION PROPERTIES





Voltage (volts)

Current-voltage characteristics in the forward (a) and reverse (b) directions for unirradiated (1) and irradiated (2) single crystal, alpha-silicon carbide p-n junctions. The original material was n-type, n $\sim 5 \times 10^{17}/cc$. Radiation used is:

I - γ-rays

II - β -rays

III - γ-rays + neutrons

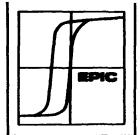
No change in current-voltage characteristics but a junction photoeffect occurs.

[Ref. 19624]

A fast neutron flux of 2.2 x $10^{17}/\text{cm}^2$ changes p-type, single crystal, α -silicon carbide to n-type. n-Type remains unaffected by a 3.8 x $10^{16}/\text{cm}^2$ flux. Optical transmission is only slightly affected by $\text{nvt}_f = 10^{16}/\text{cm}^2$ from .4 to 2.5 microns. 1 ohm-cm, n-type, α -silicon carbide, single crystal after $10^{16}/\text{cm}^2$ neutron irradiation has 100 ohm-cm resistivity.

Mobility	Sample	Irradiation	Temperature	Ref.
15 cm ² /v sec 50 cm ² /v sec	single crystal, p-type single crystal, n-type	non-irradiated nvt _f = 10 ¹⁶ /cm ²	500°K 500°K	*

^{*} CENTRE D' ETUDE DE L' ENERGIE NUCLEAIRE. MOL, BELGIUM. Solid State Neutron Detector. QR No. 6, Jan. 1 - Mar. 31, 1962. EURAEC-306. Contract No. 079-61-10RDB. 1962.



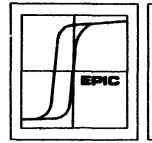
E LECTRONIC
P ROPERTIES
I NFORMATION
C ENTER

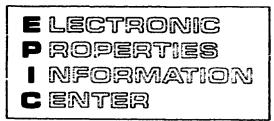
PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

LIFETIME

	Value	Sample			
Symbol	(µsec)	(single crystal)	Test Hethod I	em perature	Ref.
$^{\tau}_{\mathbf{p}}$	80	p-n junction, a-,	recomb. radiation	133°K	2152
τp	1.15	**	Ħ	300°K	2152
τp	→ 30	n-type, α-,	photoconductivity	300°K	11441
τp	1.	p-n junction, a-,	recomb. radiation	300°K	7867
τ _P	30	<pre>p-type, boron-doped, α-, ρ = 11000 ohm-cm</pre>	current voltage	300°K	5908
τp	.3	n-type, pure, argon grown, white, α -, ρ = 5300 ohm-cm	11	11	5908
τ _p	.008 to .02	n-type, nitrogen-doped, α -, ρ = .2 to 10 ohm-cm	photoconductivity & electroluminescence	330 e K	10533
τ _p	0.01	<pre>p-type, aluminum-doped, α-, to 90000 ohm-cm</pre>	н	300 ° K	10533
τ _p .01	to 1.0	n-type, aluminum-doped, α -, $n \sim 10^{18}/cc$, ρ = .5 ohm-cm	recomb. radiation after 1 Mev electron irradiation	200-350°K	15717
τ _n	3	<pre>p-type, boron-doped, α-, ρ = 11000 ohm-cm</pre>	current voltage	300 ° K	5908
$^{\tau}$ n	< 0.5	p-n junction, α -,	electroluminescence	300°K	17421
τn	∿ .4 .1	n-type, α-,	11	300 ° K	7805
	.004 .009	p-type, α-,	11	77	7805
τn	.01	p-n junction, n-type, Al-doped $n_n \sim 10^{18}/cc$	electroluminescence	300°K	368
τn	.01 to .05	n- and p-type, aluminum-doped, β -,		300 ° K	17725
τ	<3	<pre>β-, single crystal & polycrystalline 56</pre>	electrical		17416



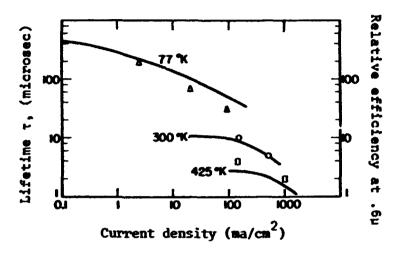


PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

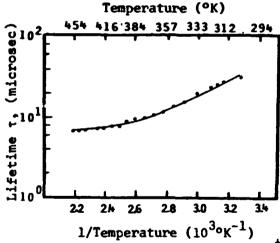
SILICON CARBIDE

LIFETIME

Lifetime as a function of current density for single crystal, alphasilicon carbide, (p-n junctions). The points give the hole lifetime and the curves give, at the same temperatures, the relative number of photons per unit current, i.e., relative efficiency, at 6 microns. The curves are normalized to bring out the correlation with the hole lifetimes.



[Ref. 368]

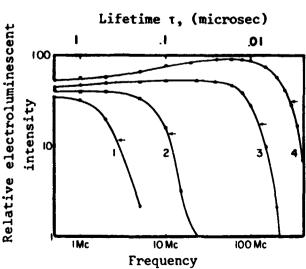


Lifetime as a function of temperature for n-type, single crystal, alphasilicon carbide; photoconductivity measurements.

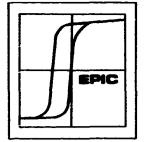
[Ref. 11441]

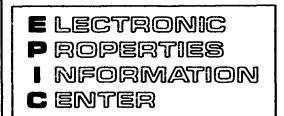
Lifetime as a function of frequency for single crystal, alpha-silicon carbide at 300°K. 1, 2 and 4 are n-type, 3 is p-type.

indicate t at 35% of maximum value.



[Ref. 7805]





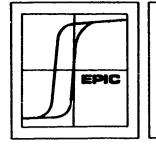
PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

MAGNETIC SUSCEPTIBILITY

Symbol	Value	Sample (single crystal)	Test Method	Temperature	Ref.
	(mean value) -12.8 x 10 ⁶ /gm mol [0.316 x 10 ⁻⁶ /cgs units]	green, α-,	4640-6160 oe	300°K	13757
x ^T	-13.1 x 10 ⁻⁶ /gm mol -0.326 x 10 ⁻⁶ /cgs units]			13757
χ _⊥	-5.4 to 8.2 x 10 ⁻⁶ /gm m	ol black, α-,			13757
x	-0.319 x 10 ⁻⁶ /cgs units				7359
	(magnetic anisotropy) 0.82 x 10 ⁻⁶ /gm mol [0.02 x 10 ⁻⁶ /cgs units]		3560-4250 oe	300 ° K	13757

Magnetic susceptibility is shown in both molar and cgs units.

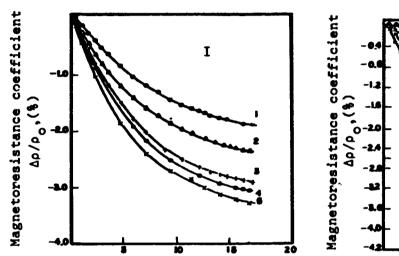


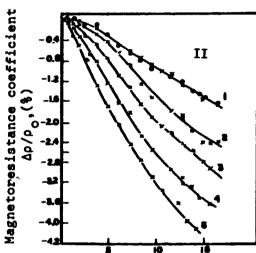


PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER + HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

. SILICON CARBIDE

MAGNETOELECTRIC PROPERTIES





Magnetic field H, (kcps)

Magnetoresistance as a function of magnetic field for fairly pure, (slightly N-doped), n-type, alpha-silicon carbide.

I)
$$n = 3 \times 10^{17}/cc$$

1 - 4.23°K

2 - 3.23 K

3 - 2.08°K

4 - 1.55°K

5 - 1.43°K

II)
$$n = 6.4 \times 10^{16}/cc$$

1 - 4.21°K

2 - 3.10°K

3 - 2.54°K

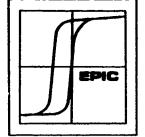
4 - 2.01°K

5 - 1.63°K

o and x are data taken with different probes.

[Ref. 17973]

AIR FORCE NATURALS LABORATORY
RESEARCH OF SYSTEMS COMMAND



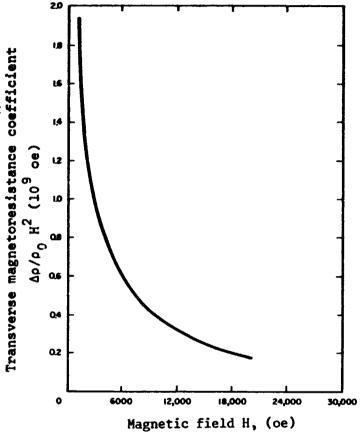
E LECTRONIC
P ROPERTIES
I NFORMATION
C ENTER

PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER • HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

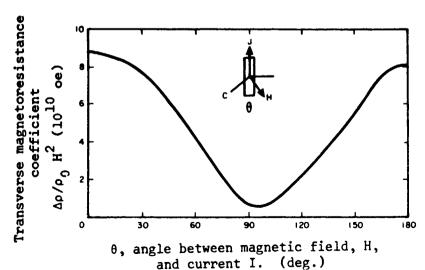
SILICON CARBIDE

MAGNETOELECTRIC PROPERTIES

Transverse magnetoresistance coefficient as a function of magnetic field in n-type, single crystal, alpha-silicon carbide at 77°K, H parallel to c-axis.

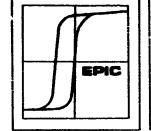


[Ref. 11132]



Transverse magnetoresistance coefficient as \vec{H} is rotated with respect to \vec{J} in n-type, single crystal, alpha-silicon carbide, at 77°K. \vec{H} = 3900 oe.

[Ref. 11132]



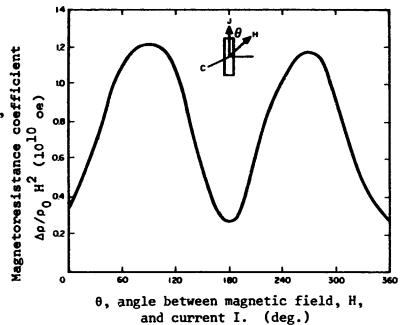


PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER • HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

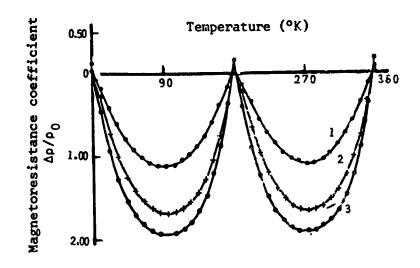
SILICON CARBIDE

MAGNETOELECTRIC PROPERTIES

Magnetoresistance as a function of angle for n-type, single crystal, alpha-silicon carbide at 77°K and 3900 oe.



[Ref. 11132]



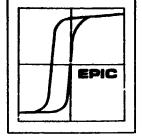
Magnetoresistance as a function of θ, angle between magnetic field and current for n-type, single crystal, alpha-silicon carbide at 4.22°K.

1 - 6.1 kc

2 - 11.6 kc

3 - 15.7 kc

[Ref. 17973]



E LECTRONIC
P ROPERTIES
I NFORMATION
C ENTER

PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

MOBILITY

Symbol	Value		Sample (s (ohm-cm)	ingle crystal)	Temperature	Ref.
μ	>2000 400	~ 10 ¹⁴	10.	n-type, pure, α-, (6H)	77°K 300°K	5464
	40 38	~ 10 ¹⁶	.3		77°K 300°K	5464
ц	360 160 142 90	2 x 10 ¹⁷ 1 30 40	.425 .2 .053	n-type, N-doped, α-, (6H)	293°K	16397
	37	200	.0085			16397
μ	26	4:	22.	p-type photo. meas.	at 300°K	409
μ _n	2100 3265 4670	1.67 x 10 ¹⁶ 1.18 x 10 ¹⁶ 1.38 x 10 ¹⁵	.179 .163 .971	n-type, very pure, β-,	300°K	17725 17725
^μ n	150 62	1.3 x 10 ¹⁷ 3.6 x 10 ¹⁷		n-type, N-doped, α-,	300°K 500°K	17233 17233
μ _n	32		.00185	n-type, light green, β-,	300°K	17416
μ η p	220 48	< 10 ¹⁵		n-type, pure, α -, (argon grow	n) 300°K	8783 8783

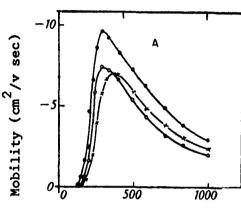
EPIC



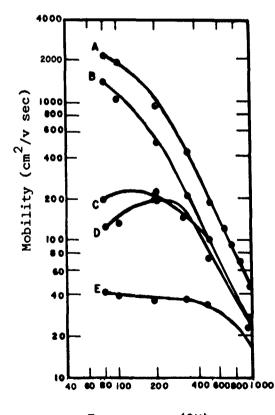
PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

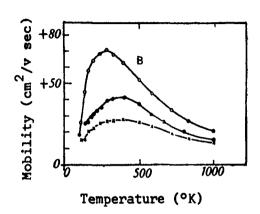
MOBILITY



Temperature (°K)



Temperature (°K)



[Ref. 5990]

Mobility as a function of temperature for single crystal, alpha-silicon carbide.

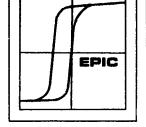
A - p-type, black, $\rho_{1000K} \sim .05$ ohm-cm.

B - n-type, green,
$$\rho = \begin{pmatrix} 0 \\ x \\ 0 \end{pmatrix}$$
 .12 at 480°K 620°K 570°K

Mobility as a function of temperature for very pure, n-type, single crystal, alphasilicon carbide, (6H).

[Ref. 5464]

AIR FORCE MATERIALS LABORATORY RESEARCH AND TECHNOLOGY DIVISION





PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

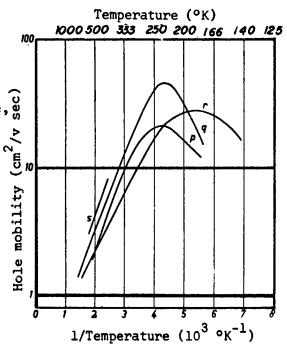
SILICON CARBIDE

MOBILITY

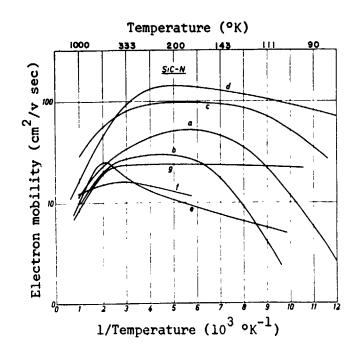
Mobility as a function of temperature for p-type, single crystal, alpha-silicon carbide, prepared in argon, argon + boron chloride, and argon + aluminum chloride.

p - undoped
$$n \sim 10^{17}/cc$$

q - Al-doped $n = 2.4 \times 10^{18}$
- B-doped $n = 2.3 \times 10^{17}$
r - Al-doped $n = 3.7 \times 10^{18}$
n = 9.5 x 10¹⁷
s - Al-doped $n = 10^{18}$



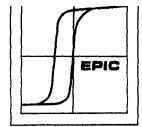
[Ref. 3608]



Mobility as a function of temperature for n-type, single crystal, alphasilicon carbide, prepared in argon, or argon + nitrogen.

a - no dopant
$$n_n = 3.6 \times 10^{17}/cc$$

b - N-, A1-, B-, $n \sim 10^{17}$
c - N-, A1-, B-, $n \sim 10^{18}$
d - N-, A1-, B-, $n = 3 \times 10^{18}$
e - N-, A1-, B-, $n = 10^{19}$
f - N-doped, $n = 5 \times 10^{19}$
g - N-doped $n = 3.4 \times 10^{19}$

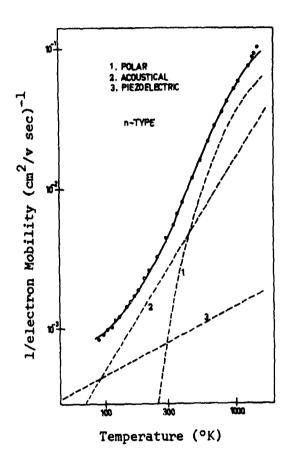


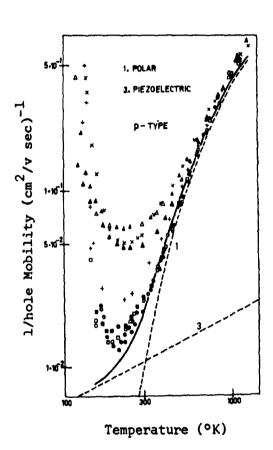
E LECTRONIC
P ROPERTIES
I NFORMATION
C ENTER

PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

MOBILITY



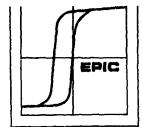


Mobility as a function of temperature for p-type, single crystal, alpha-silicon carbide, (6H). Principal dopants are aluminum, boron, and chlorine. Nitrogen is always present. Dashed lines are calculated for - 1) polar, 2) acoustical, and 3) piezoelectric scattering. The solid line is the sum of the two and the dots show experimental points.

- argon grown (fairly pure)
- toxygen, oxygen and chlorine
- △ { aluminum
- boron
- o chlorine-doped

[Ref. 8783]

AIR FORCE MATERIALS LABORATORY RESEARCH AND TECHNOLOGY DIVISION





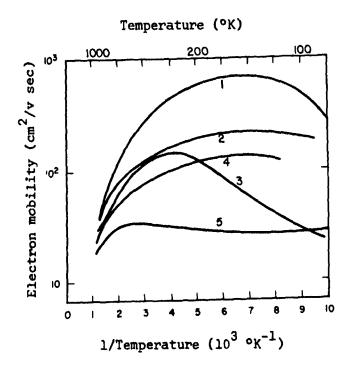
PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

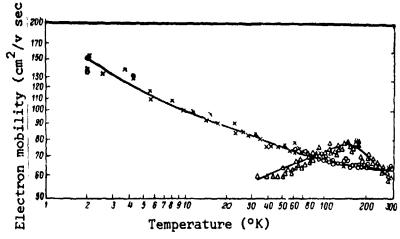
MOBILITY

Mobility as a function of temperature for n-type, N-doped, single crystal, alpha-silicon carbide.

^р 300К	ohm-cm	n /cc
	0.425	10 ¹⁷
2 -	0.2	10 ¹⁷
3 -	0.53	3×10^{18}
4 -	0.48	4 x 10 ¹⁸
5 -	0.0085	2×10^{19}



[Ref. 16397]

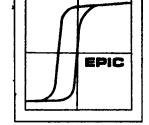


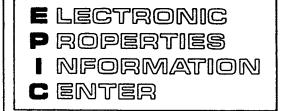
Electron mobility as a function of temperature for n-type, single crystal, alpha-silicon carbide, (lll) oriented.

n /ce ρ_{290K} ohm-cm

1) o 5.7 x 10¹⁸ .0175
2) x .0174
3) ∇ 5.6 x 10¹⁷ .222

[Ref. 17973]

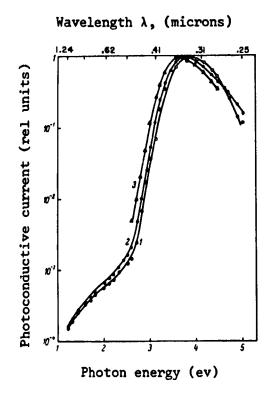




PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

PHOTOELECTRONIC PROPERTIES

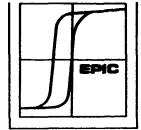


[Ref. 4596]

Photoconductivity as a function of wavelength for n-type, single crystal, alpha-silicon carbide, (6H), (001) oriented.

$$\rho_{300K}$$
 = 7.5 ohm-cm.

- 1 100°K
- 2 293°K
- 3 500°K





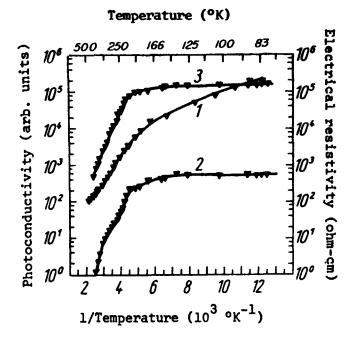
PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

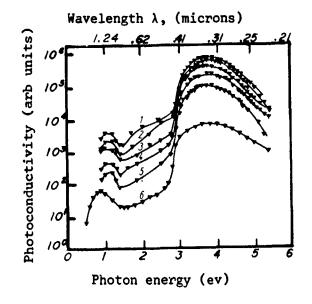
PHOTOELECTRONIC PROPERTIES

Photoconductivity and electrical resistivity as a function of temperature for n-type, single crystal, alpha-silicon carbide.

- 1 electrical resistivity
- 2 photoconductivity at 1.03 microns
- 3 photoconductivity at 0.32 microns



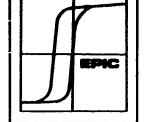
[Ref. 10994]

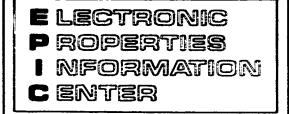


Photoconductivity as a function of wavelength for n-type, single crystal, alphasilicon carbide.

- 1 125°K
- 2 180°K
- 3 200°K
- $4 230^{\circ}K$
- 5 260°K
- 6 300°K

[Ref. 10994]





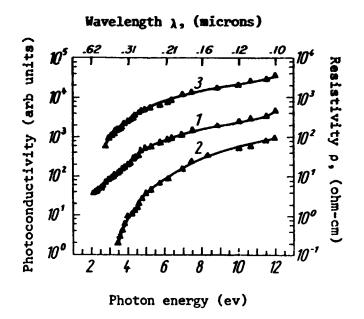
PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

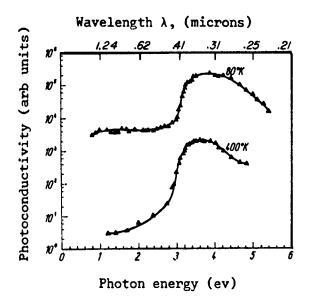
PHOTOELECTRONIC PROPERTIES

Photoconductivity and electrical resistivity as a function of temperature for p-type, single crystal, alpha-silicon carbide.

- 1 electrical resistivity
- 2 photoconductivity at 1.05 microns
- 3 photoconductivity at 0.35 microns

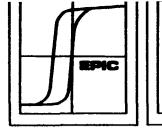


[Ref. 10994]



Photoconductivity as a function of wavelength for p-type, single crystal, alphasilicon carbide at 80°K and 400°K.

[Ref. 10994]



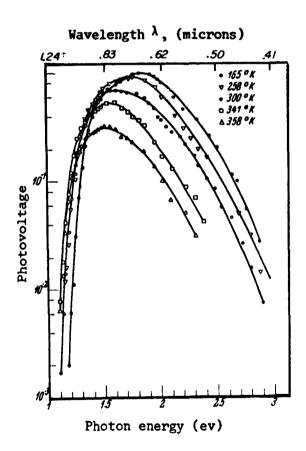


PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

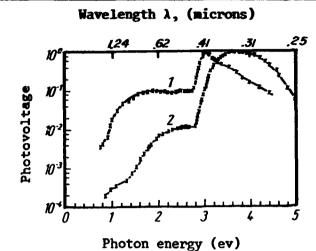
SILICON CARBIDE

PHOTOELECTRONIC PROPERTIES

Photoconductivity (1) and photovoltage (2) for n-type, single crystal, alpha-silicon carbide at 300°K. Photovoltage normalized to 1. Photoconductivity placed only for comparison, without units.



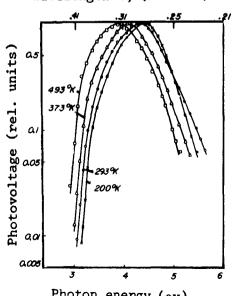
Photovoltage as a function of wavelength for p-n junction, single crystal alpha-silicon carbide at various temperatures.



[Ref. 10994]

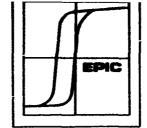
Photovoltage as a function of wavelength for p-type, single crystal, alpha-silicon carbide at several temperatures. Photovoltage normalized to 1.

[Ref. 7741] Wavelength λ , (microns)



Photon energy (ev) [Ref. 7867]

AR PORCE MATERIALS LABORATORY RESERVOIR FORCE SYSTEMS COMMAND



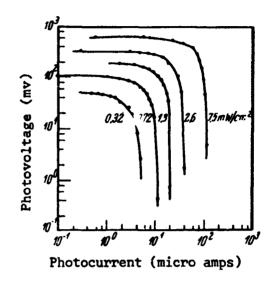


PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

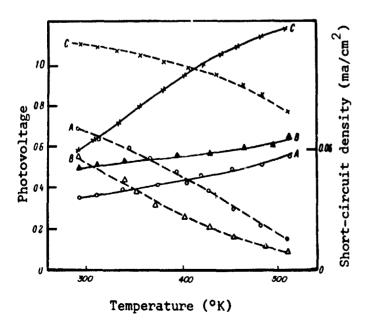
. SILICON CARBIDE

PHOTOELECTRONIC PROPERTIES

Current-Voltage curve of an n-type, single crystal, alpha-silicon carbide photoelement at light intensities from .32 to 7.5 mW/cm².



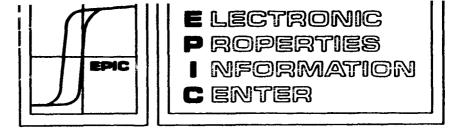
[Ref. 7741]



Photovoltage and short circuit current density as a function of temperature for p-n junction, single crystal, alpha-silicon carbide at constant illumination.

- short circuit density
- --- photovoltage

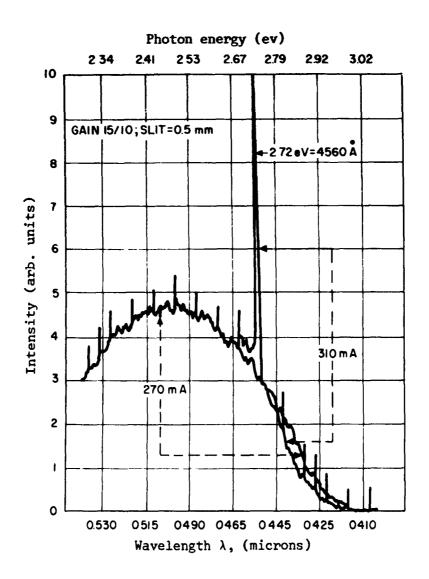
[Ref. 7867]



PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

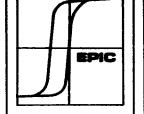
SILICON CARBIDE

PHOTON ELECTROLUMINESCENCE



Stimulated emission in a forward-biased p-n junction of alpha-silicon carbide. Emission peak at .456 microns has a linewidth <0.005 microns. Continuous operation at 300°K around 120 amps/cm².

[Ref. 17972]



E LECTRONIC
P ROPERTIES
I NFORMATION
C ENTER

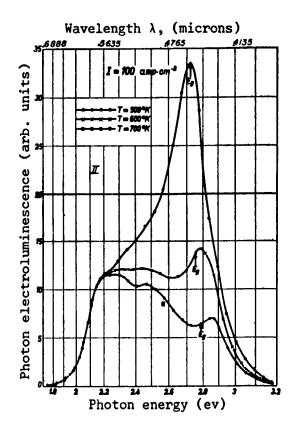
PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

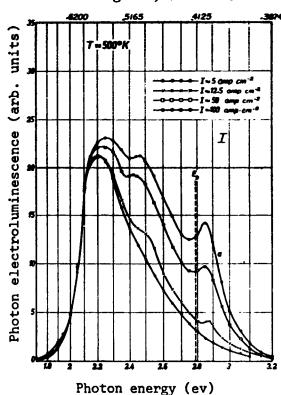
PHOTON ELECTROLUMINESCENCE

Photo electroluminescence as a function of wavelength for single crystal, alpha-silicon carbide p-n junction at 500°K, and several current intensities.

a is the same curve on both graphs on a different scale.



Wavelength λ , (microns)

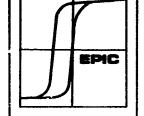


[Ref. 18406]

Photo electroluminescence as a function of wavelength for single crystal, alpha-silicon carbide p-n junction at several temperatures and 100 amps/cm².

[Ref. 18406]

AIR FORCE NATERIALS LABORATORY RESEARCH AND TECHNOLOGY DIVISION



E LECTRONIC
P ROPERTIES
I NFORMATION
C ENTER

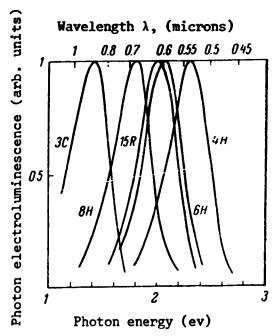
PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

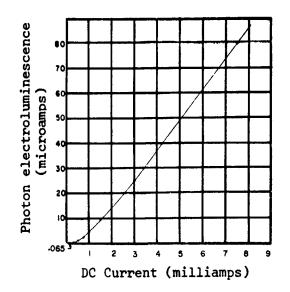
PHOTON ELECTROLUMINESCENCE

Electroluminescence as a function of wavelength in single crystal, alpha-silicon carbide p-n junctions of several polytypes at 300° K. ρ < .3 ohm-cm. The 3C curve is for n-type, beta-silicon carbide, ρ < 0.4 ohm-cm. Current density was 0.05 to 0.1 amp/cm².

Photon electroluminescence is measured in photons/unit time and energy.



[Ref. 18818]



Electroluminescence as a function of DC current in fairly pure, light green, single crystal, alpha-silicon carbide p-n junction at 300°K.

The curve starts at .065 milliamp.

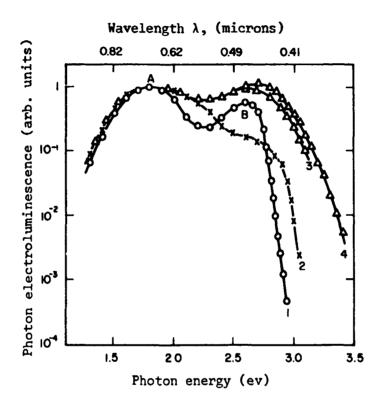
[Ref. 2152]

AIR FORCE NATERIALS LABORATORY RESEARCH AND STATEMENT OF STATEMENT OF

PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICUN CARBIDE

PHOTON ELECTROLUMINESCENCE



Photon electroluminescence in single crystal, alpha-silicon carbide p-n junction. Various impurities are present.

$$n \sim 10^{18}/cc.$$
, $\rho = .2$ ohm-cm.

 $1 - 130^{\circ}K$, 1 amp/cm^2

 $2 - 400^{\circ}K$, 1 amp/cm^2

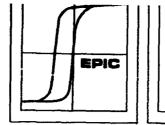
3 - 700°K, 1 amp/cm^2

4 - 700°K, 100 amps/cm^2

[Ref. 17457]

Photon electroluminescence is measured in photons/unit time and energy.

AIR PORCE MATERIALS LABORATORY
RESEARCH AND TECHNOLOGY DIVISION



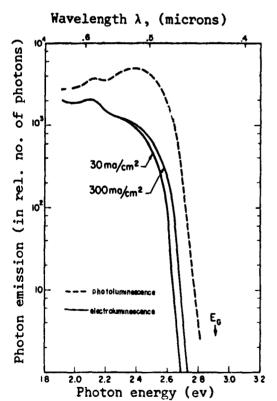


PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

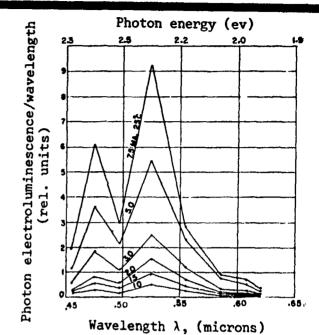
SILICON CARBIDE

PHOTON ELECTROLUMINESCENCE

Normalized photon electroluminescence as a function of wavelength in n-type, single crystal, pale green, alpha-silicon carbide at 300°K. The current varies from 1.0 to 7.5 microamps.



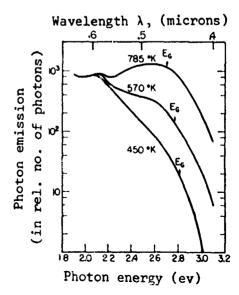
Electroluminescence as a function of wavelength for single crystal, alpha-silicon carbide p-n junction, at several temperatures and 1 amp/cm². Curves are normalized at .6 microns.



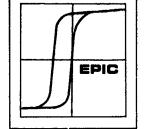
[Ref. 2152]

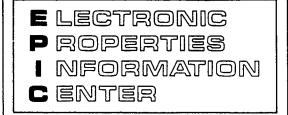
Photon and electroluminescence as a function of wavelength for single crystal, alpha-silicon carbide p-n junction at 77°K. Increase in excitation by a factor of 10 is indicated in electroluminescent spectra by shift to higher wavelength, but similar increase in UV excitation has no effect on photoluminescent spectrum.

[Ref. 368]



[Ref. 368]

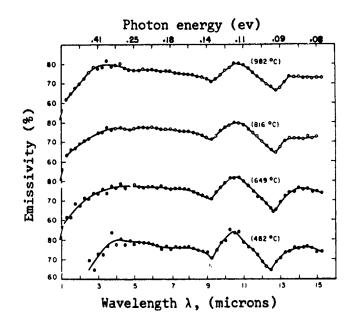




PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

PHOTON EMISSIVITY

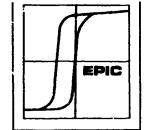


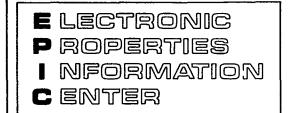
[Ref. 2217]

Spectral emissivity of Globar heating element as a function of wavelength at four temperatures.

Spectral Emissivity

Value	Sample	Temperature	Ref.
0.85%	sintered silicon carbide rods at 1.6 microns	1243°K	17724
0.93%	polycrystalline commercial silicon carbide	1500°F	1956



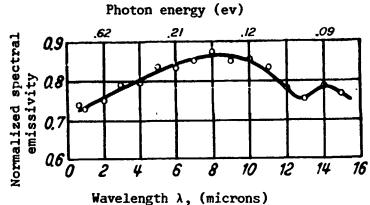


PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

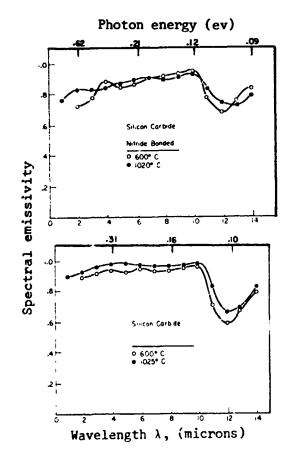
SILICON CARBIDE

PHOTON EMISSIVITY

Spectral emissivity as a function of wavelength for sintered silicon carbide rods at 1243°K. Spectral emissivity is normalized to the standard black body base.

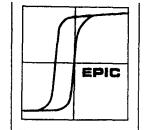


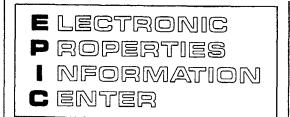
[Ref. 17724]



Spectral emissivity as a function of wavelength for commercially pure, fine grained silicon carbide at 2 temperatures.

[Ref. 17412]





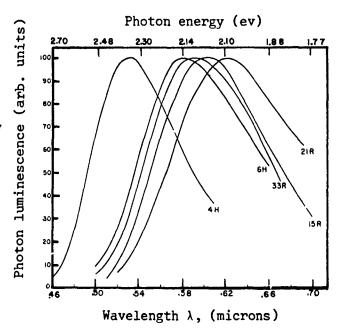
PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

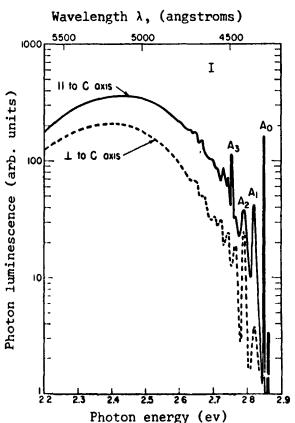
PHOTON LUMINESCENCE

Photon luminescence as a function of wavelength for alpha-silicon carbide powder, B-, and N-doped, at 300°K.

Curves indicate the polytype from x-ray determinations.



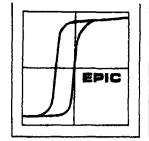
[Ref. 17404]

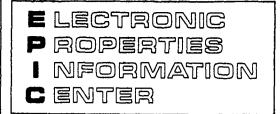


Photon luminescence as a function of wavelength for single crystal, alphasilicon carbide, (6H), at 4°K, (001) oriented. The subscripts indicate the level number.

[Ref. 11690]

AIR FORCE MATERIALS LABORATORY
RESEARCH AND TECHNOLOGY DIVISION

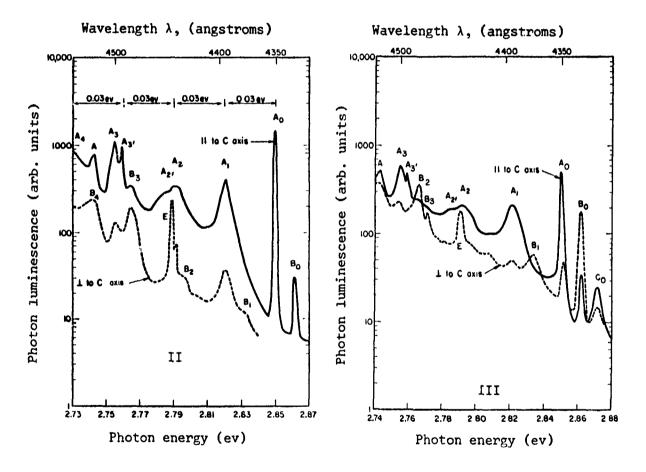




PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER • HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

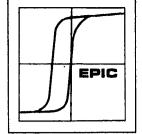
SILICON CARBIDE

PHOTON LUMINESCENCE



[Ref. 11690]

Photon luminescence for single crystal, alpha-silicon carbide, (6H), at 4°K and 77°K, giving fine structure of the peaks in I.





PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

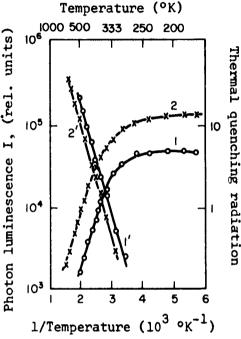
SILICON CARBIDE

PHOTON LUMINESCENCE

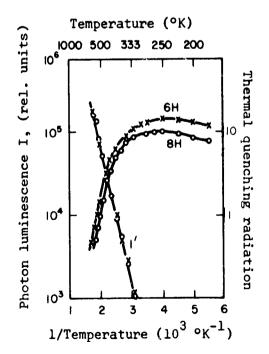
Photon luminescence as a function of temperature for single crystal, beta-silicon carbide p-n junction, (nitrogen-boron). $\rho = 0.08-0.4$ ohm-cm.

1, 2 - photoluminescence for UV irradiation

1', 2', - quenching curves whose slope equal thermal quenching activation energy 0.29-0.3 ev.



[Ref. 18818]



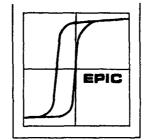
Photon luminescence as a function of temperature for single crystal, alphasilicon carbide (6H) and (8H) p-n junction, (nitrogen-boron). p = 0.05 to 0.3 ohm-cm.

1' is quenching curve whose slope is \sim 0.35 ev.

[Ref. 18818]

AIR FORCE MATERIALS LABORATORY

RESEARCH AND TECHNOLOGY DIVISION AIR FORCE SYSTEMS COMMAND





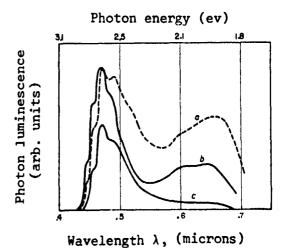
PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

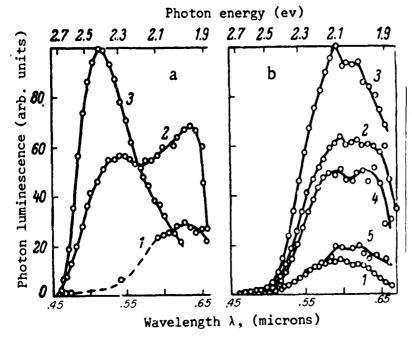
PHOTON LUMINESCENCE

Photon luminescence as a function of wavelength for single crystal, alphasilicon carbide at 63°K.

- a n-type, N-doped, white, $n \sim 10^{18}/cc$
- b n-type, N-doped, green, n = 2.7×10^{18}
- c p-type, Al-doped, light blue, $n \sim 10^{18}$



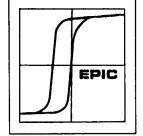
[Ref. 3607]



Photon luminescence as a function of wavelength for single crystal, alpha-silicon carbide, (6H), at several temperatures.

- a) transparent colorless crystal,
 - 1 423°K
 - 2 123°K
 - 3 90°K
- b) black crystal,
 - 1 423°K
 - 2 333°K
 - 3 273°K
 - 4 194°K
 - 5 90°K

[Ref. 13404]



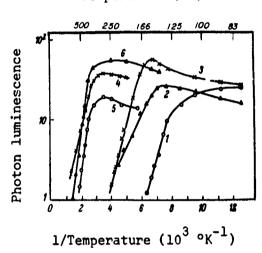
E LECTRONIC P ROPERTIES NFORMATION C ENTER

PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

PHOTON LUMINESCENCE

Temperature (°K)



curves 1, 2, 3

Photon luminescence as a function of temperature for n-type, single crystal, alpha-silicon carbide, (6H), nitrogen-doped.

$$n = 10^{17}$$
 to 5 x $10^{18}/cc$, $\rho_{300K} = 0.02$ to 1 ohm-cm.

curves 4, 5, 6

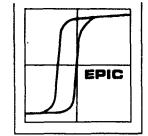
Photon luminescence as a function of temperature for single crystal, alpha-silicon carbide p-n junctions.

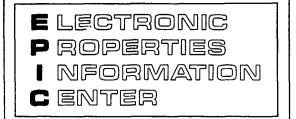
> Curves 4 and 5 are for the (6H) polytype Curve 6 is for the (4H) polytype.

> > [Ref. 17730]

AIR FORCE MATERIALS LABORATORY

RESEARCH AND TECHNOLOGY DIVISION AIR FORCE SYSTEMS COMMAND





PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

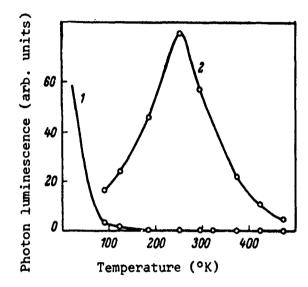
SILICON CARBIDE

PHOTON LUMINESCENCE

Photon luminescence as a function of temperature for n-type, single crystal, alpha-silicon carbide, (6H).

1 - blue luminescence

2 - orange



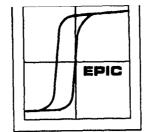
[Ref. 13404]

Photoluminescence maxima, dopant, and electrical resistivity for single crystal, beta-silicon carbide.

Crystal Type	Dopant	Resistivity (ohm-cm)	Luminescence at 77°K I	Cuminescence at 300°K Color
n p-n p n n n n	A1 A1 A1 +N ₂ + A1 A1, Mo + N ₂ A1, + N ₂ Mg	10 ³ 10 10-1 10-2 10-1 1 - 10 1 - 10 4 - 10 10-1	brilliant orange brilliant pink brilliant pink bright orange bright pink orange pink orange dull orange very dull green, pink, yellow	yellow none none dull red yellow dull red none green, yellow, red yellow, red, green none
n n	Ca B	1 - 10	very dull red very dull orange	dull red bright yellow, red & green

[Ref. 17725]

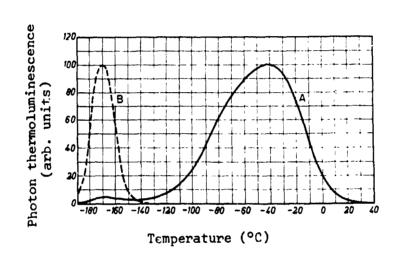
AIR FORCE MATERIALS LABORATORY
RESEARCH AND TECHNOLOGY DIVISION



PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

PHOTON THERMOLUMINESCENCE



Photon thermoluminescence as a function of temperature for single crystal, alpha-silicon carbide. Excitation at 83°K and .365 microns.

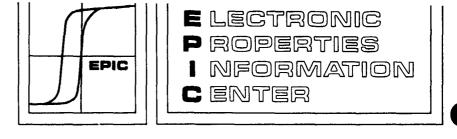
A - p-type, Al-doped, light blue, n $\sim 10^{18}/cc$.

B - n-type, nitrogen-doped, white, n $\sim 10^{18}/cc$.

[Ref. 3607]

AIR FORCE MATERIALS LABORATORY

RESEARCH AND TECHNOLOGY DIVISION AIR FORCE SYSTEMS COMMAND



PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

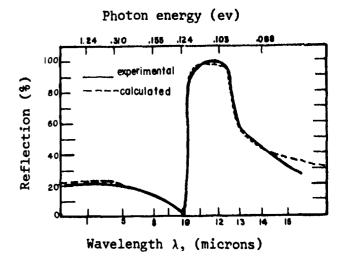
REFLECTIVITY

Reflectivity as a function of wavelength for pure, single crystal, alpha-silicon carbide at 300°K.

E || to the c-axis

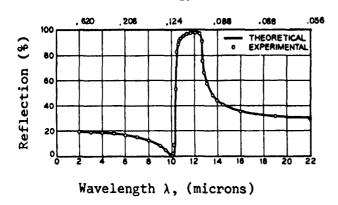
n_1 to the c-axis

 n_{ω} | to the c-axis



[Ref. 3607]

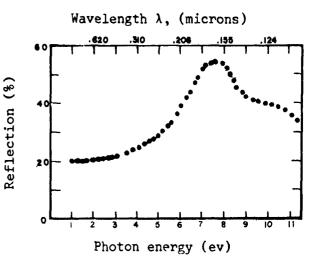
Photon energy (ev)



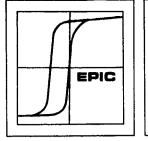
Resistivity as a function of wavelength for single crystal, alphasilicon carbide, (6H), at 300°K. E is parallel to the c-axis.

[Ref. 628]

Reflectivity as a function of wavelength for single crystal, alpha-silicon carbide at 300°K. Below 3 ev these values are calculated from direct measurements of index of refraction in [Ref. 13873].



[Ref. 17419]





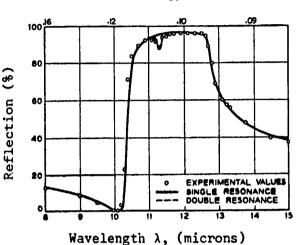
PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

REFLECTIVITY

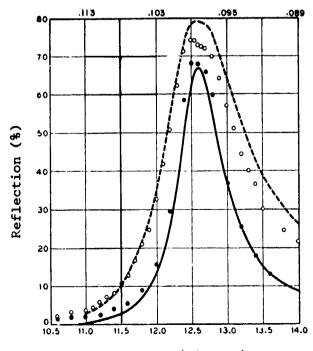
Reflectivity as a function of wavelength for single crystal, alpha-silicon carbide, (6H), at 300°K. E is normal to the c-axis.

Photon energy (ev)



[Ref. 628]

Photon energy (ev)

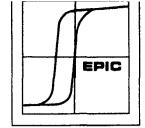


Wavelength λ , (microns)

Reflectivity as a function of wavelength for polycrystalline, betasilicon carbide films at 300°K.

- - 0.06 microns thick
- o 0.12 microns thick

[Ref. 627]

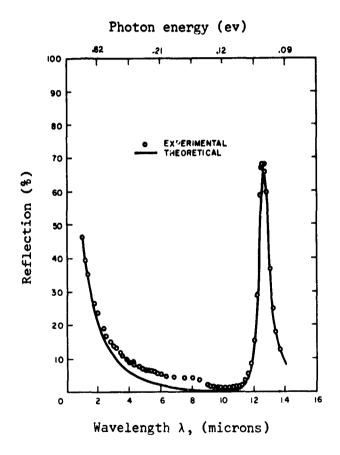


E LECTRONIC
P ROPERTIES
I NFORMATION
C ENTER

PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

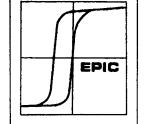
SILICON CARBIDE

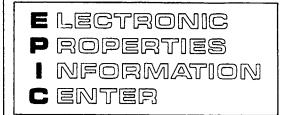
REFLECTIVITY



[Ref. 627]

Reflectivity as a function of wavelength for polycrystalline, beta-silicon carbide film, 0.06 microns thick.





PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

REFRACTIVE INDEX

Symbol	Value (at	300°K)		Test Method welength)	Ref.
n	∿ 2.63		β -, yellow to olive green	.6708µ	13873
	ε 2.673 2.697 2.721	ω 2.633 2.654 2.675	α-, (001), green-black (Moissanite)	.6708µ .5895µ .5351µ	*Larsen
	2.6889	2.6475	α-, (6H), light green	.5895µ	13873
	2.6892	2.6468	α-, (15R), pale green	.5895µ	13873
	2.654	2.616	α-, pale green	.755µ	**Gmelin

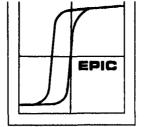
^{.6708}µ is Li

^{.5895}µ is Na

^{.5351}µ is T1

^{*} Larsen, E.S., and H. Berman. The Microscopic Determination of the Nonopaque Minerals. U.S.G.P.O., Washington, 1934. p. 77

^{**} GMELINS HANDBUCH DER ANORGANISCHEN CHEMIE; achte völlig neu bearbeitete Auflage. Silicium. Teil B. Weinheim, Verlag Chemie, GMBH, 1959. p. 827.





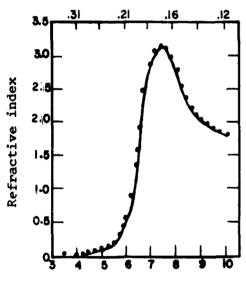
PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

REFRACTIVE INDEX

Index of refraction of single crystal, alphasilicon carbide, (6H), at 300°K. Below 3 ev the values are taken from [Ref. 13873].

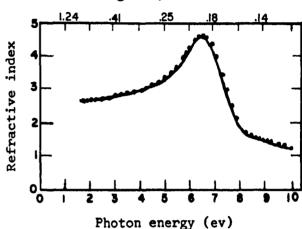
Wavelength λ , (microns)



Photon energy (ev)

Refractive index as a function of wavelength for pure, single crystal, alphasilicon carbide at 300°K. (Calculated from reflectivity data.)

Wavelength λ , (microns)

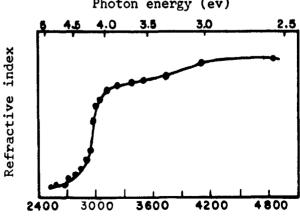


[Ref. 17419]

The extinction coefficient, i.e., the imaginary part of the index of refraction, of single crystal, alpha-silicon carbide, (6H), at 300°K.

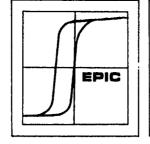
[Ref. 17419]

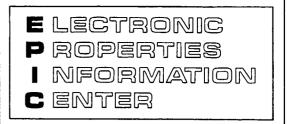
Photon energy (ev)



Wavelength λ , (angstroms)

[Ref. 13404]



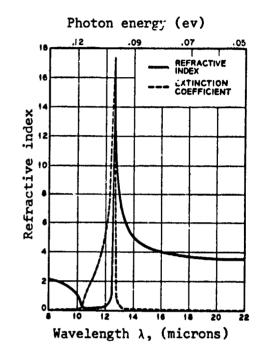


PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

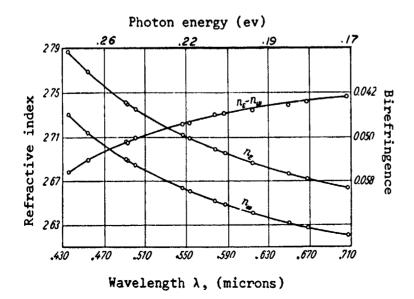
SILICON CARBIDE

REFRACTIVE INDEX

The theoretically calculated optical constants as a function of wavelength for single crystal, n-type, alpha-silicon carbide, (6H).

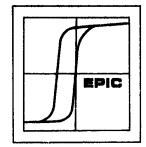


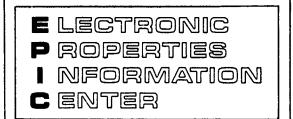
[Ref. 628]



Refractive index of ordinary n_{ϵ} rays and extraordinary n_{ω} rays as a function of wavelength in fairly pure, single crystal, alpha-silicon carbide at 300°K. n_{ϵ} - n_{ω} is the birefringence curve.*

AIR FORCE MATERIALS LABORATORY RESEARCH AND TECHNOLOGY DIVISION





PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

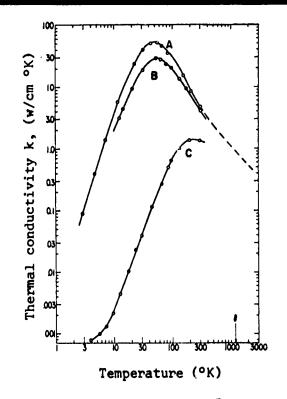
THERMAL CONDUCTIVITY

Thermal conductivity as a function of temperature for single crystal, alphasilicon carbide of varied purity.

A - n-type, N-doped, colorless, (6H), $n \sim 10^{17}/cc.$, $\rho_{300K} \sim 10$ ohm-cm.

B - n-type, N- & Al-doped, light green, (6H), $n \sim 10^{19}/cc.$, $\rho_{300K} = 2$ ohm-cm.

C - p-type, Al-doped, dark blue, (6H & 15R), $n = 4 \times 10^{19}/cc.$, $\rho_{300K} \sim .6$ ohm-cm.



[Ref. 17406]

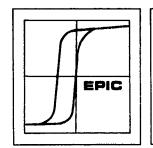
Thermal conductivity of sample A. Slightly N-doped.

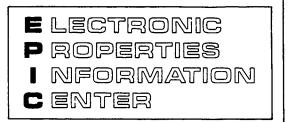
Temperature °K	Thermal Conductivity w/cm deg.	Ref.	
50	52	17406	
70	43		
100	28		
150	15		
200	9.5		
300	4.9		
500	2.4*	,	
1000	1.1*		
1200	0.9*	17406	
1478	.157**	1956	
343	2.05***	17374	

^{*} extrapolated

*** polycrystalline

^{**} commercial polycrystalline, alpha-silicon carbide





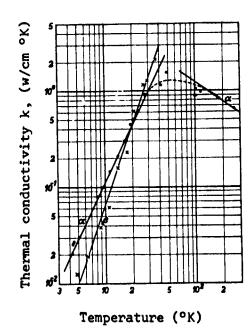
PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

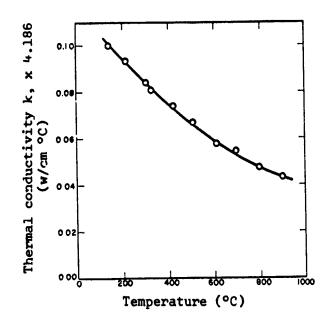
THERMAL CONDUCTIVITY

Thermal conductivity as a function of temperature for single crystal, alphasilicon carbide, (6H), and N-doped, polycrystalline, beta-silicon carbide.

$$n < 10^{17}/cc.$$

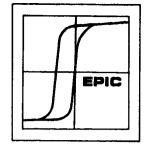


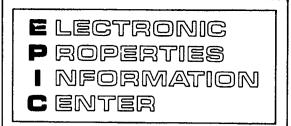
[Ref. 19150]



Thermal conductivity as a function of temperature for commercial, polycrystalline, alpha-silicon carbide.

AIR FORCE MATERIALS LABORATORY RESEARCH AND TECHNOLOGY DIVISION



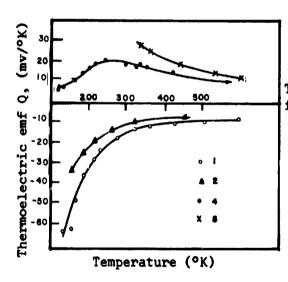


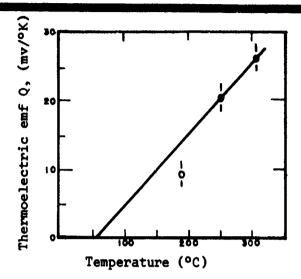
PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

THERMOELECTRIC PROPERTIES

Thermoelectric emf as a function of temperature for n-type, single crystal, yellow, beta-silicon carbide. $\rho \sim 1$ to 100 ohm-cm.





[Ref. 17416]

Thermoelectric emf as a function of temperature for single crystal, alpha-silicon carbide with resistivities to 10⁴ ohm-cm.

$$1 - n$$
-type, $n = 5.6 \times 10^{16}/cc$

$$2 - n$$
-type, $n = 2.7 \times 10^{17}$

$$4 - n$$
-type, $n = 2.7 \times 10^{17}$

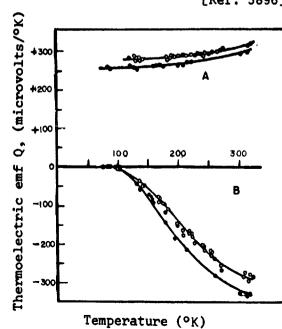
$$5 - p-type, n = 2.0 \times 10^{18}$$

[Ref. 5896]

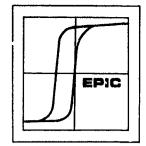
Thermal emf as a function of temperature for single crystal, alpha-silicon carbide.

A - n-type, N-doped, green

B - p-type, Al-doped, black



[Ref. 6031]

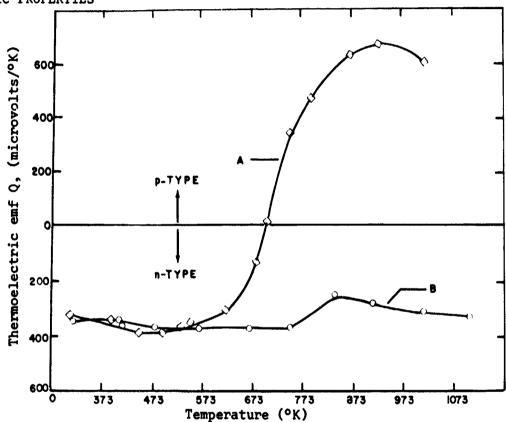




PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

THERMOELECTRIC PROPERTIES

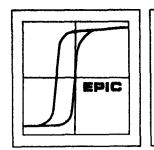


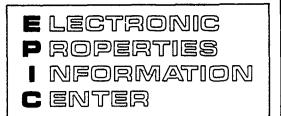
Thermoelectric emf as a function of temperature for two silicon carbide samples.

- A Single crystal, alpha-silicon carbide, ρ = 45 ohm-cm, colorless, becomes p-type above 680°K.
- B Single crystal, n-type, alpha-silicon carbide, ρ = 0.026 ohm-cm, dark green, remains n-type with heating.

[Ref. 19560]

•	Symbol	Value (µv/°K)	Sample	Ref.
•	Q	20-50	polycrystalline commercial material, $\rho \sim 0.05$ ohm-cm	3974
	•	200 ± 10% -106	polycrystalline, β-SiC rods n-type, single crystal, yellow, β-,	17416
			$\rho = 0.1$ to 100 ohm-cm	17416





PREPARED BY ELECTRONIC PROPERTIES INFORMATION CENTER . HUGHES AIRCRAFT COMPANY, CULVER CITY, CALIFORNIA

SILICON CARBIDE

WORK FUNCTION

Symbol '	Value (ev)	Sample (single crystal)	Test Method	Temperature	Ref.
ф	4.62	<pre>impure, p-type, α-, ρ = 0.3 ohm-cm., black</pre>	photoelectric emission	300°K 	7134
ф	4.55	impure, p-type, α -, ρ = 0.6 ohm-cm., green	11		7134
ф	n 4	impure, α-,	electron thermionic emission	1000-1700°K	7152
	ev .3 to 4)	impure, n- and p-type, black green to light gr	reen yellow	300-800°K	2468
	ron affinity ψ	β-	photoelectric data	300°K	787
	electric thresh	old β-	**	**	787

REFERENCES

- 368 LYLE, P. and W.J. CHOYKE. Impurity Bands and Electroluminescence in SiC p-n Junctions. J. OF APPLIED PHYS., v. 30, no. 2, Feb. 1959. p. 236-248.
- 409 VAN DER MAESEN, F. Determination of Numbers of Injected Holes and Electrons in Semiconductors. PHILIPS RES. REPTS., v. 15, no. 2, Apr. 1960. p. 107-119.
- 627 SPITZER, W.G., et al. Infrared Properties of Cubic Silicon Carbide Films. PHYS. REV., v. 113, no. 1, Jan. 1, 1959. p. 133-136.
- 628 SPITZER, W.G., et al. Infrared Properties of Hexagonal Silicon Carbide. PHYS. REV., v. 113, no. 1, Jan. 1, 1959. p. 127-132.
- 787 PHILLIP, H.R. Intrinsic Optical Absorption in Single-Crystal Silicon Carbide. PHYS. REV., v. 111, no. 2, July 15, 1958. p. 440-441.
- 953 PRIMAK, W., et al. Radiation Damage in Diamond and Silicon Carbide. PHYS. REV., v. 103, no. 5, Sept. 1, 1956. p. 1184-1192.
- 1956 BROWN, R.W. Taking the Black Magic Out of High-Temperature Nonmetallics. CHEM. ENG., v. 65, Apr. 21, 1958. p. 146-147.
- 2152 LEHOVEC, K., et al. Light Emission Produced by Current Injected into a Green Silicon-Carbide Crystal. PHYS. REV., v. 89, no. 1, Jan. 1, 1953. p. 20-25.
- 2217 STEWART, J.E. and J.C. RICHMOND. Infrared Emission Spectrum of Silicon Carbide Heating Elements. NAT. BUREAU OF STANDARDS, J. OF RES., v. 59, no. 6, Dec. 1957. p. 405-409.
- 2348 KOBAYASI, S. Calculation of the Energy Band Structure of the Beta-SiC Crystal by the Orthogonalized Plane Wave Method. PHYS. SOC. OF JAPAN, J., v. 13, no. 3, Mar. 1958. p. 261-268.
- 2419 LEHOVEC, K. and H. KEDESDY. Graphical Determination of the Fermi Level in a Simple Impurity Semiconductor. J. OF APPLIED PHYS., v. 22, no. 1, Jan. 1951. p. 65-67.
- 2468 HOLM, E. Contribution to the Theory of the Silicon Carbide Contact. J. OF APPLIED PHYS., v. 23, no. 5, May 1952. p. 509-517.
- 2514 RUECKER, D. Zur Elektrolumineszenz des Siliziumkarbids. The Electroluminescence of Silicon Carbide. ZEITSCHRIFT FUER ANGEWANDTE PHYS., v. 10, no. 6, 1958. p. 254-263.
- 2863 NAMBA, M. Optical Properties of Silicon Carbide. PHYS. AND CHEM. OF SOLIDS, v. 2, no. 4, 1957. p. 339-340.
- 2899 SZIGETI, G. Ueber die Durch Gleichstrom Angeregte Elektrolumineszenz von Siliziumkarbidkristallen. The D.C. Activated Electroluminescence of Silicon Carbide Crystals. ACTA PHYSICA HUNGARICA, v. 4, no. 1, 1954. p. 65-77.
- 3394 DALVEN, R. Absorption Edge in Beta-Silicon Carbide. FHYS. AND CHEM. OF SOLIDS., v. 13, no. 1-2, May 1960. p. 163-164.

- 3607 LELY, J.A. and F.A. KROEGER. Optical Properties of Pure and Doped SiC. In: SEMICONDUCTORS AND PHOSPHORS. HALBLEITER UND PHOSPHORE. PROC. OF INTERNAT. COLLOQUIUM-PARTENKIRCHEN. Ed. by SCHOEN, M. and H. WELKER. N.Y., Intersci. Pub., Inc. 1958. p. 514-524.
- 3608 LELY, J.A. and F.A. KROEGER. Electrical Properties of Hexagonal SiC Doped with N, B, or Al. In: SEMICONDUCTORS AND PHOSPHORS. HALBLEITER UND PHOSPHORE. PROC. OF INTERNAT. COLLOQUIUM-PARTENKIRCHEN. Ed. by SCHOEN, M. and H. WELKER. N.Y., Intersci. Pub., Inc., '958. p. 525-533.
- 3974 NATIONAL CARBON CO., INC. Thermoelectric Materials, by THIELKE, N.R. Bimonthly progress rept. no. 2, Mar. 28-May 28, 1959. June 15, 1959. Contract NObs-77066. ASTIA AD-229 057.
- 4434 GROTH, R. and E. KAUER. Absorption freier Ladungstraeger in alpha-SiC-Kristallen. Absorption by Free Charge Carriers in alpha-SiC Crystals. PHYSICA STATUS SOLIDI, v. 1, no. 5, 1961. p. 445-450.
- 4596 POROZHNIKOVA, G.N. and G.F. KHOLUYANOV. Photoconductivity in alpha-Silicon Carbide. SOVIET PHYS. SOLID STATE, v. 3, no. 12, June 1962. p. 2657-2661.
- WESTINGHOUSE ELECTRIC CORP. Research in the Preparation of Hyperpure Single Crystal Silicon Carbide, by HAMILTON, D.R., et al. Final Rept. Feb. 1957-June 1959. Apr. 1960. AFCRL TR 60-368. Contract AF 19 604 2174. ASTIA AD-250 119.
- 5896 LOMAKINA, G.A. and Yu. A. VODAKOV. Effect of Phonon Drag in alpha-SiC Crystals. SOVIET PHYS. SOLID STATE, v. 4, no. 3, Sept. 1962. p. 603-604.
- 5908 OZAROW, V. and R.E. HYSELL. Space-Charge-Limited Currents in Single-Crystal Silicon Carbide. J. OF APPLIED PHYS., v. 33, no. 10, Oct. 1962. p. 3013-3015.
- 5990 BUSCH, G. and H. LABHART. Ueber den Mechanismus der elektrischen Leitfaehigkeit des Siliziumkarbids. On the Mechanism of the Electrical Conductivity in Silicon Carbide. HELVETICA PHYSICA ACTA, v. 19, no. 6-7, 1946. p. 463-492.
- BUSCH, G.. et al. Thermospannungen des Siliziumkarbids gegen Kupfer und Platin. Thermo-Vo cages of Silicon Carbide with Copper or Platinum. HELVETICA PHYSICA ACTA, v. 27, no. 6, 1947. p. 461-463.
- 6080 BUSCH, G. Die elektrische Leitfaehigkeit des Siliziumkarbids. Electrical Conductivity of Silicon Carbide. HELVETICA PHYSICA ACTA, v. 19, no. 3, 1946. p. 167-188.
- 6397 WOOLEY, J.C. and B. RAY. Effects of Solid Solution of In Te 3 with A II B Tellurides. PHYS. AND CHEM. OF SOLIDS, v. 15, no. 1-2, Aug. 1960. p. 27-32.
- 6505 CHOYKE, W.J. and L. PATRICK. Exciton Recombination Radiation and Phonon Spectrum of 6H SiC. PHYS. REV., v. 127, no. 6, Sept. 15, 1962. p. 1868-1877.
- 7134 DILLON, J.A., JR., et al. Some Surface Properties of Silicon-Carbide Crystals, J. OF APPLIED PHYS., v. 30, no. 5, May 1959. p. 675-679.

- 7152 BRAUN, A. and G. BUSCH. Versuche zur Messung der thermischen Elektronenemission von Graphit, Silizium und Siliziumkarbid. Investigations in Measurements of Thermal Electron Emission of Graphite, Silicon and Silicon Carbide. HELVETICA PHYSICA ACTA, v. 20, no. 1, Feb. 15, 1947. p. 33-66.
- 7333 PATRICK, L. Structure and Characteristics of Silicon Carbide Light-Emitting Junctions. J. OF APPLIED PHYS., v. 28, no. 7, July 1957. p. 765-776.
- 7359 AIGRAIN, P. and M. BALKANSKI. Selected Constants Relative to Semi-Conductors. Oxford, N.Y., Pergamon Press, 1961.
- 7741 THIESSEN, K. and G. JUNGK. Photo-EMK bei Stoerstellen-und Grundgitteranregung im SiC. Photo-EMF for Defect Points and Basic Lattice Excitation in SiC. PHYSICA STATUS SOLIDI, v. 2, no. 4, 1962. p. 473-478.
- 7805 HARMAN, G.G. and R.L. RAYBOLD. Measurement of Minority Carrier Lifetime in SiC by a Novel Electroluminescent Method. J. OF APPLIED PHYS., v. 32, no. 6, June 1961. p. 1168-1169.
- 7867 KHOLUYANOV, G.F. Photoelectric Properties of Alloy p-n Junctions in Silicon Carbide. SOVIET PHYS. SOLID STATE, v. 2, no. 8, Feb. 1961. p. 1722-1726.
- 8783 VAN DAAL, H.J., et al. On the Electronic Conduction of alpha-SiC Crystals between 300 and 1500°K. J. OF PHYS. AND CHEM. OF SOLIDS, v. 24, Jan. 1963. p. 109-127.
- 8792 CHOYKE, W.J. and L. PATRICK. Exciton and Interband Absorption in SiC. In: INTERNAT. CONF. ON SEMICONDUCTOR PHYS., PROC., PRAGUE, 1960. N.Y. Acad. Press, 1961. p. 432-434.
- 10533 KHOLUYANOV, G.F. Photoconductivity Relaxation and the Charge Carrier Lifetime in Silicon Carbide. SOVIET PHYS. SOLID STATE, v. 4, no. 11, May 1963. p. 2322-2326.
- 10994 JUNGK, G., et al. Stationare Photoleitung und p-n-Photoeffekt in SiC. Stationary Photoconductivity and p-n Photoeffect in SiC. PHYSICA STATUS SOLIDI, v. 3, no. 4, 1963. p. 735-743.
- 11132 MASS. INST. OF TECH. LINCOLN LAB. Solid State Research. QPR, Aug. 1, 1957. ASTIA AD-151 261.
- 11441 THIESSEN, K., et al. Zur Kinetik der Photoleitung in SiC. The Kinetics of Photoconductivity in SiC. PHYSICA STATUS SOLIDI, v. 3, no. 3, 1963. p. 529-534.
- 11690 CHOYKE, W.J., et al. Polarized Edge Emission of SiC. PHYS. REV., v. 117, no. 6, Mar. 15, 1960. p. 1430-1438.
- 11938 SZIGETI, G. Short Lecture Given During the Session on Electroluminescence. BRITISH J. OF APPLIED PHYS., v. 6, supp. no. 4, 1955. p. s56.
- 13156 PATRICK, L., et al. Optical Properties of 15R Silicon Carbide Luminescence of Nitrogen-Exciton Complexes, and Interband Absorption. PHYS. REV., v. 132, no. 5, Dec. 1, 1963. p. 2023-2031.

- 13404 GORBAN, I.S. and S.N. RUDKO. Optical Properties of Silicon Carbide Crystals. SOVIET PHYS. SOLID STATE, v. 5, no. 5, Nov. 1963. p. 995-998.
- 13757 SIGAMONY, A. The Magnetic Susceptibility and Anisotropy of Carborundum. INDIAN ACAD. OF SCI., PROC., A, v. 19, 1944. p. 377-380.
- 13873 THIBAULT, N.W. Morphological and Structural Crystallography and Optical Properties of Silicon Carbide. AMERICAN MINERALOGIST, v. 29, no. 9-10, Sept.-Oct., 1944. p. 327-362.
- 14587 BASSANI, F. and M. YOSHIMINE. Electronic Band Structure of Group IV Elements and of III-V Compounds. PHYS. REV., v. 130, no. 1, Apr. 1, 1963. p. 20-33.
- 14663 STEPANOV, G.V., et al. Emission of Hot Electrons from Natural p-n Junctions in Silicon Carbide Crystals. SOVIET PHYS. SOLID STATE, v.3, no. 6, Dec. 1961. p. 1280-1283.
- 15717 VAN DER DOES DE BYE, J.A.W. Transient Recombination in Silicon Carbide. PHILIPS RES. REPTS., v. 17, no. 5, Oct. 1962. p. 418-430.
- 16397 VIOLINA, G.N., et al. Optical Absorption and Electrical Properties of n-Type alpha-Silicon Carbide. SOVIET PHYS. SOLID STATE, v. 5, no. 12, June 1964. p. 2500-2505.
- 17233 1MAI, A. Free Carrier Absorption in n-Type Silicon Carbide. PHYS. SOC. OF JAPAN, J., v. 19, no. 8, Aug. 1964. p. 1486.
- 17374 WEEKS, J.L. and R.L. SEIFERT. Apparatus for the Measurement of the Thermal Conductivity of Solids. REV. OF SCI. INSTRUMENTS, v. 24, no. 11, Nov. 1953. p. 1054-1057.
- 17403 CARROLL, P. Resistivity of Granular Silicon Carbide Impurity and Crystal Structure Dependence. In: CONF. ON SILICON CARBIDE, BOSTON, 1959. SILICON CARBIDE, A HIGH TEMPERATURE SEMICONDUCTOR, PROC., Ed. by O'CONNOR, J.R. and J. SMILTENS. N.Y., Symposium Pub. Div., Pergamon Press, 1960. p. 341-346.
- 174C+ ADDAMIANO, A. Photoluminescence and Polytypism in Boron Doped Silicon Carbide. ELECTROCHEM. SOC., J., v. 111, no. 11, Nov. 1964. p. 1294.
- 17405 CHOYKE, W.J., et al. Optical Properties of Cubic Silicon Carbide. Luminescence of Nitrogen-Exciton Complexes, and Interband Absorption. PHYS. REV., v. 133, no. 4A, Feb. 17, 1964. p. All63-All66.
- 17406 SLACK, G.A. Thermal Conductivity of Pure and Impure Silicon Carbide and Diamond. J. OF APPLIED PHYS., v. 35, no. 12, Dec. 1964. p. 3460-3466.
- 17411 VASILOS, T. and W.D. KINGERY. Thermal Conductivity: XI, Conductivity of Some Refractory Carbides and Nitrides. AMERICAN CERAM. SOC., J., v. 37, no. 9, Sept. 1954. p. 409-414.
- 17412 BLAU, H.H., JR. and J.R. JASPERSE. Spectral Emittance of Refractory Materials. APPLIED OPTICS, v. 3, no. 2, Feb. 1964. p. 281-286.
- 17414 FRIDEL, I. and G.F. KHOLUYANOV. Electron Emission from p-n Junctions in Silicon Carbide Prepared by Nitrogen Diffusion. SOVIET PHYS. SOLID STATE, v. 5, no. 7, Jan. 1964. p. 1418-1421.

- 17415 LIPSON, H.G. Infra-Red Transmission of alpha-Silicon Carbide. In: CONF. ON SILICON CARBIDE, BOSTON, 1959. SILICON CARBIDE, A HIGH TEMPERATURE SEMI-CONDUCTOR, PROC., Ed. by O'CONNOR, J.R. and J. SMILTENS. N.Y., Symposium Pub. Div., Pergamon Press, 1960. p. 371-375.
- 17416 POHL, R.G. Electrical Properties of beta-Silicon Carbide. In: CONF. ON SILICON CARBIDE, BOSTON, 1959. SILICON CARBIDE, A HIGH TEMPERATURE SEMI-CONDUCTOR, PROC., ED. by O'CONNOR, J.R. and J. SMILTENS. N.Y., Symposium Pub. Div., Pergamon Press, 1960. p. 312-330.
- 17417 CHOYKE, W.J. and L. PATRICK. Absorption of Light in alpha-Silicon Carbide Near the Band Edge. In: CONF. ON SILICON CARBIDE, BOSTON, 1959. SILICON CARBIDE, A HIGH TEMPERATURE SEMICONDUCTOR, PROC., Ed. by O'CONNOR, J.R. and J. SMILTENS. N.Y., Symposium Pub. Div., Pergamon Press, 1960. p. 306-311.
- 17419 PHILIPP, H.R. and E.A. TAFT. Intrinsic Optical Absorption in Single Crystal Silicon Carbide. In: CONF. ON SILICON CARBIDE, BOSTON, 1959. SILICON CARBIDE, A HIGH TEMPERATURE SEMICONDUCTOR, PROC., Ed. by O'CONNOR, J.R. and J. SMILTENS. N.Y., Symposium Pub. Div., Pergamon Press, 1960. p. 366-371.
- 17420 SPITZER, W.G., et al. Infra-Red Properties of Silicon Carbide. In: CONF. ON SILICON CARBIDE, BOSTON, 1959. SILICON CARBIDE, A HIGH TEMPERATURE SEMICONDUCTOR, PROC., Ed. by O'CONNOR, J.R. and J. SMILTENS. N.Y., Symposium Pub. Div., Pergamon Press, 1960. p. 376-382.
- 17421 ERIKSEN, W.T. Minority Carrier Lifetime in Silicon Carbide by Carrier Injection Electroluminescence. In: CONF. ON SILICON CARBIDE, BOSTON, 1959. SILICON CARBIDE, A HIGH TEMPERATURE SEMICONDUCTOR, PROC., Ed. by O'CONNOR, J.R. and J. SMILTENS. N.Y., Symposium Pub. Div., Pergamon Press, 1960. p. 376-332.
- 17457 TOMASHPOLSKII, F.G. and G.F. KHOLUYANOV. Concerning Recombination Radiation Spectra from Fused p-n Junctions in Sic. SOVIET PHYS. SOLID STATE, v. 3, no. 9, Mar. 1962. p. 1960-1963.
- 17724 BRUEGEL, W. Strahlungsmessungen an elektrisch geheizten SiC-Staeben. Radiation Measurements on Electrically Heated Silicon Carbide Rods. ZEITSCHRIFT FUER PHYSIK, v. 127, no. 3, Mar. 17, 1950. p. 400-404.
- 17725 STANFORD RES. INST. Beta-Silicon Carbide and its Potential for Devices, by NELSON, W.E., et al. Final Rept. no. 21. Contract no. NObsr-87235. Dec. 31, 1963. DDC AD-428 005.
- 17730 VIOLIN, E.E. and G.F. KHOLUYANOV. On the Electroluminescence and Photoluminescence of Diffused p-n Junctions in SiC. SOVIET PHYS. - SOLID STATE, v. 6, no. 6, Dec. 1964. p. 1331-1335.
- 17972 GRIFFITHS, L.B., et al. Silicon Carbide Diode Laser. IEEE PROC., v. 51, no. 10, Oct. 1963. p. 1374-1376.
- 17973 IGLITSYN, M.I., et al. Galvanomagnetic Effects in n-Type Silicon Carbide at Low Temperatures. SOVIET PHYS. SOLID STATE, v. 6, no. 9, Mar. 1965. p. 2129-2135.

- 17974 PATRICK, L., et al. Luminescence of 4H Silicon Carbide, and Location of Conduction-Band Minima in Silicon Carbide Polytypes. PHYS. REV., v. 137, no. 5A, Mar. 1, 1965. p. A1515-A1520.
- 18056 DALVEN, R. Temperature Coefficient of the Energy Gap of beta-Silicon Carbide.
 J. OF PHYS. AND CHEM. OF SOLIDS, v. 26, no. 2, Feb. 1965. p. 439-441.
- 18349 LEUSHKINA, G.V., et al. Fluorescence of Silicon Carbide. AKADEMIIA NAUK UZBEKSKOI SSR. IZVESTIIA. SERIIA FIZIKO-MATEMATICHESKIKH NAUK, v. 7, no. 6, 1963. p. 98-99.
- 18406 VAN DAAL, H.J., et al. Investigations on Silicon Carbide. J. OF APPLIED PHYS., v. 32, no. 10, Oct. 1961. p. 2225-2233.
- 18818 KHOLUYANOV, G.F. Polytypism and Recombination Radiation of p-n Junctions Prepared by the Diffusion of Boron in n-type Silicon Carbide. SOVIET PHYS. SOLID STATE, v. 6, no. 11, May 1965. p. 2668-2671.
- 19150 BOSCH, G. On the Thermal Conductivity of Silicon Carbide. PHILIPS RES. REPTS., v. 16, no. 5, Oct. 1961. p. 455-461.
- 19560 NAT. BUREAU OF STANDARDS. DEPT. OF COMMERCE. Electroluminescence and Thermoelectricity, by HARMAN, G.G. and R.L. RAYBOLD. TR no. 1, AFCRC-TN-59-999. Dec. 1, 1959. DDC AD-220 814.

PUBLICATIONS OF THE ELECTRONIC PROPERTIES INFORMATION CENTER (EPIC)

Data Sheets

- DS-101 Cadmium Telluride. M. Neuberger. June 1962. (AD-415 331)
- DS-102 Indium Phosphide. M. Neuberger. June 1962. (AD-414 847)
- DS-103 Indium Telluride. M. Neuberger. June 1962. (AD-414 896)
- DS-104 Magnesium S.licide. M. Neuberger. June 1962. (AD-414 895)
- DS-105 Polyethylene Terephthalate. J.T. Milek. June 1962. (AD-414 846)
- DS-106 Polytetrafluoroethylene Plastics. E. Schafer. June 1962. (AD-413 907)
- DS-107 Polytrifluorochloroethylene Plastics. E. Schafer. June 1962. (AD-413 940)
- DS-108 Zinc Telluride. M. Neuberger. June 1962. (AD-413 939)
- DS-109 Indium Arsenide. M. Neuberger. June 1962. (AD-413 892)
- DS-110 Aluminum Antimonide. M. Neuberger. September 1962. (AD-413 676)
- DS-111 Gallium Phosphide. M. Neuberger. September 1962. (AD-413 954)
- DS-112 Gallium Antimonide. M. Neuberger. October 1962. (AD-413 775)
- DS-113 Lead Telluride. M. Neuberger. October 1962. (AD-437 311)
- DS-114 Magnesium Stannide. M. Neuberger. November 1962. (AD-413 825)
- DS-115 (No longer available superseded by DS-144)
- DS-116 Lead Selenide. M. Neuberger. December 1962. (AD-437 310)
- DS-117 (No longer available superseded by DS-137)
- DS-118 (No longer available superseded by DS-137)
- DS-119 (No longer available superseded by DS-137)
- DS-120 (No longer available superseded by DS-137)
- DS-121 Indium Antimonide. M. Neuberger. February 1963. (AD-413 754)
- DS-122 Steatite. J.T. Milek. February 1963. (AD-413 834)
- DS-123 Beryllium Oxide. J.T. Milek. March 1963. (AD-413-831)
- DS-124 Cadmium Sulfide. Summary Review and Data Sheets. M. Neuberger. April 1963. (AD-413 667)
- DS-125 Magnesium Oxide. J.T. Milek. June 1963. (AD-413 809)

- DS-126 (No longer available superseded by DS-137)
- DS-127 Silicone Rubber. J.T. Milek. June 1963. (AD-413 906)
- DS-128 Cordierite. J.T. Milek. June 1963. (AD-413 850)
- DS-129 Forsterite, J.T. Milek, August 1963. (AD-421 829)
- DS-130 Pyroceram. J.T. Milek. August 1963. (AD-421 883)
- DS-131 (No longer available superseded by DS-143)
- DS-132 Zinc Selenide. M. Neuberger. September 1963. (AD-421 964)
- DS-133 Zinc Oxide. M. Neuberger. October 1963. (AD-425 212)
- DS-134 Cadmium Selenide. M. Neuberger. November 1963. (AD-425 216)
- DS-135 Zinc Sulfide. M. Neuberger and D.L. Grigsby. December 1963. (AD-427 288)
- DS-136 Aluminum Oxide. J.T. Milek. March 1964. (AD-434 173)
- DS-137 Silicon. M. Neuberger. May 1964. (AD-601 788)
- DS-138 Borosilicate Glasses. J.T. Milek. June 1964. (AD-602 773)
- DS-139 Aluminosilicate Glasses. J.T. Milek. July 1964. (AD-444 101)
- DS-140 Sulfur Hexafluoride. J.T. Milek. October 1964. (AD-607 949)
- DS-141 Niobium. D.L. Grigsby. November 1964. (AD-608 398)
- DS-142 Fluorocarbon Gases. J.T. Milek. November 1964. (AD-608 897)
- DS-143 Germanium. M. Neuberger. February 1965. (AD-610 828)
- DS-144 Gallium Arsenide. M. Neuberger. April 1965.
- DS-145 Silicon Carbide. M. Neuberger. June 1965.

Other Publications

- EPIC BULLETIN. v. 1, no. 1, January 1965 (published monthly)
- 5171.2/8 Information Retrieval Program. Electronic/Electrical Properties of Materials. First Quarterly Progress Report. E.M. Wallace. October 10, 1961. (AD-270 500)
- 5171.2/17 Information Retrieval Program. Electronic/Electrical Properties of Materials. Second Quarterly Progress Report. E.M. Wallace. January 15, 1962. (AD-271 926)

5171.2/32 Information Retrieval Program. Electronic/Electrical Properties of Materials. Third Quarterly Progress Report. E.M. Wallace. April 15, 1962. (AD-275 439) P62-18 Electrical and Electronic Properties of Materials Information and Retrieval Program. Final Report. H. Thayne Johnson, E. Schafer and E.M. Wallace. June 1962. (AD-289 546) S-3 A Survey Materials Report on Tetrafluoroethylene (TFE) Plastics. J.T. Milek. September 1964. (AD-607 798) S-4 The Purpose and Functions of the Electronic Properties Information Center. H. Thayne Johnson. August 1964. Electron Mobility in Aliphatic Hydrocarbons as Related to Organic S-5 Insulation Breakdown. J.T. Milek. February 1965. S-6 Optical Properties and Thermal Conductivity of Aluminum Oxide. M. Neuberger. February 1965.

AD numbers are furnished, when available, to facilitate your requesting publications from Defense Documentation Center (DDC), Cameron Station, Alexandria, Virginia 22314.